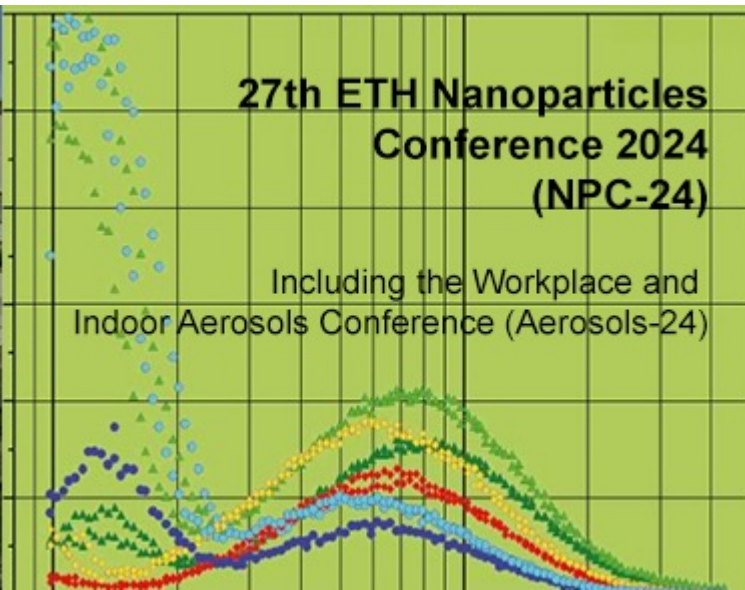


The effect of hydrogen addition on combustion generated nanoparticles from premixed ethylene flames



Stijn van Rijn, Haiyan Ni, Martijn Goudberg, Merel
 van Helten, Anatoli Mokhov, Ulrike Dusek

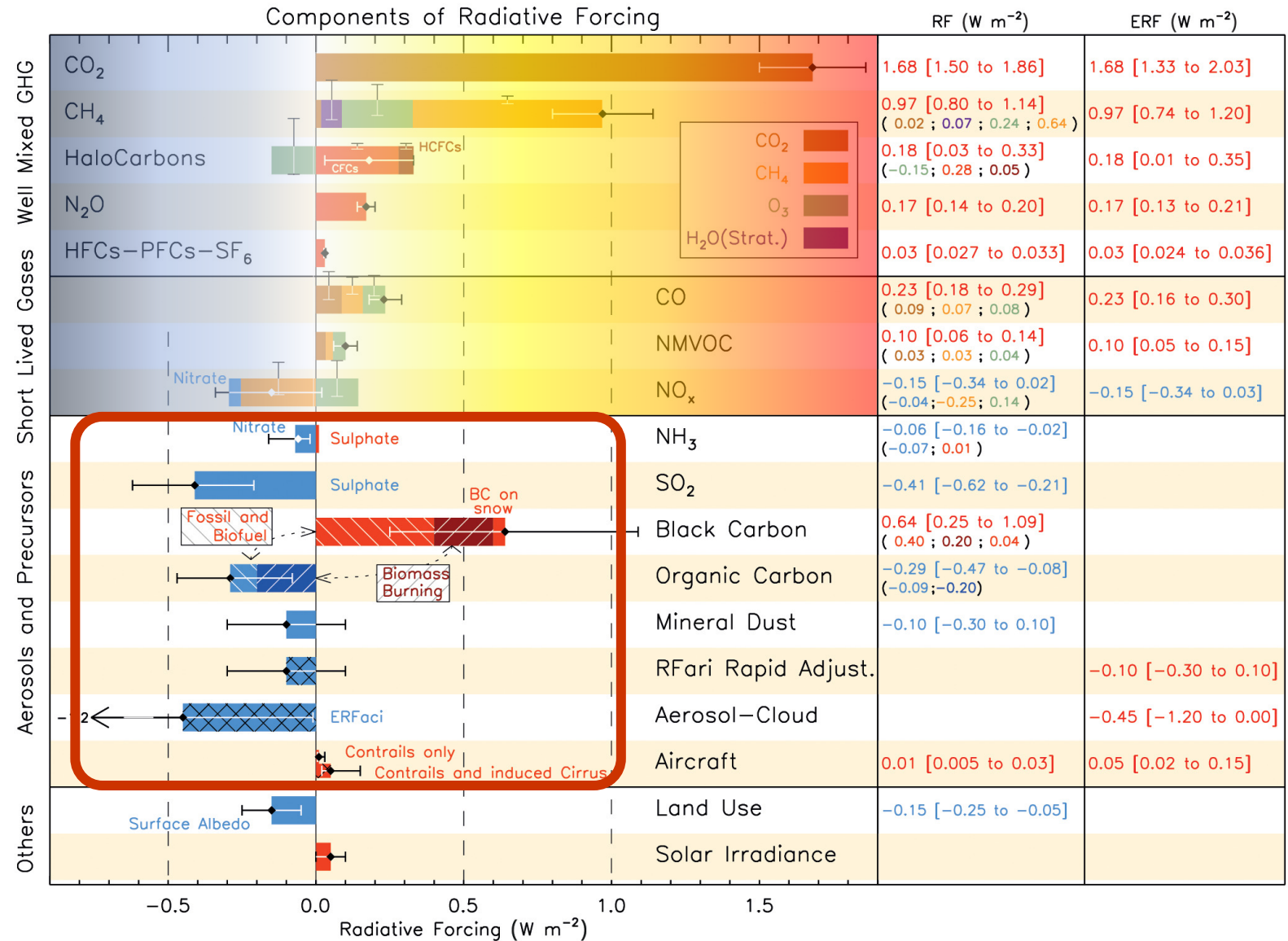
Date: 12-06-2024





Motivation

- Environmental impact of aerosols
- Black carbon
- Very potent climate forcer
- Very short lived

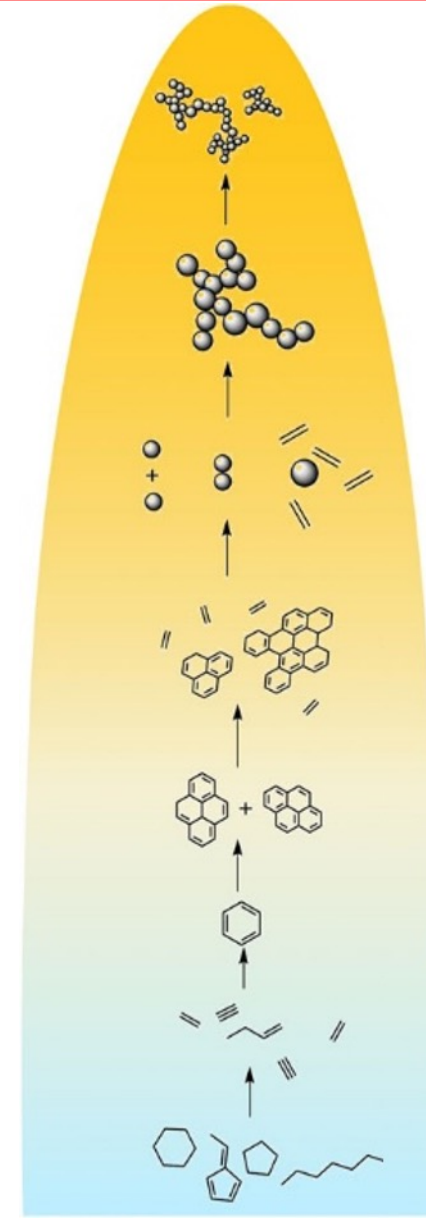


Different types of carbon aerosols



- Black Carbon
- Organic Carbon
- Brown Carbon
- Elemental Carbon
- Total Carbon
- Comparison is hard

	Thermochemical Classification	Molecular Structure	Optical Classification	
↑ Refractiveness	Elemental Carbon (EC)	Graphene Layers (graphitic or turbostratic)	Black Carbon (BC)	↑ Specific Absorption
	Refractory Organics	Polycyclic Aromatics, Humic-Like Substances, Biopolymers, etc.	Colored Organics (BrC)	
	Non-Refractory Organics	Low-MW hydrocarbons and Derivatives (carboxylic acids, etc.)	Colorless Organics	

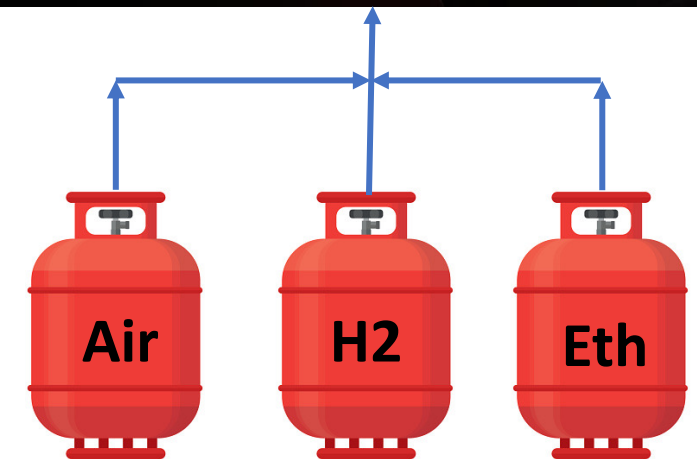
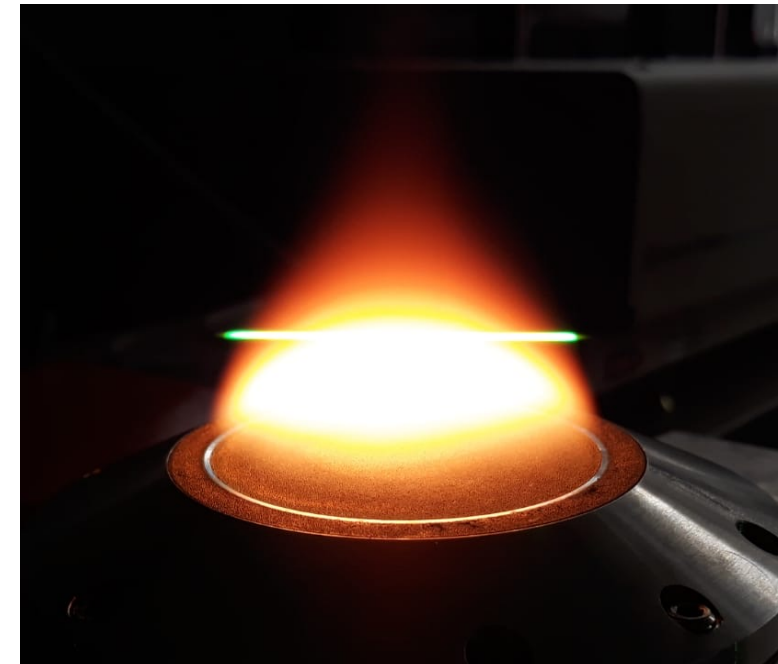
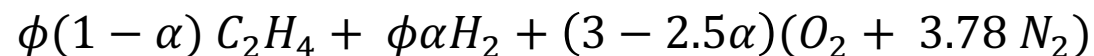


- Formation of soot
- Might co-fueling be a (temporary) solution?



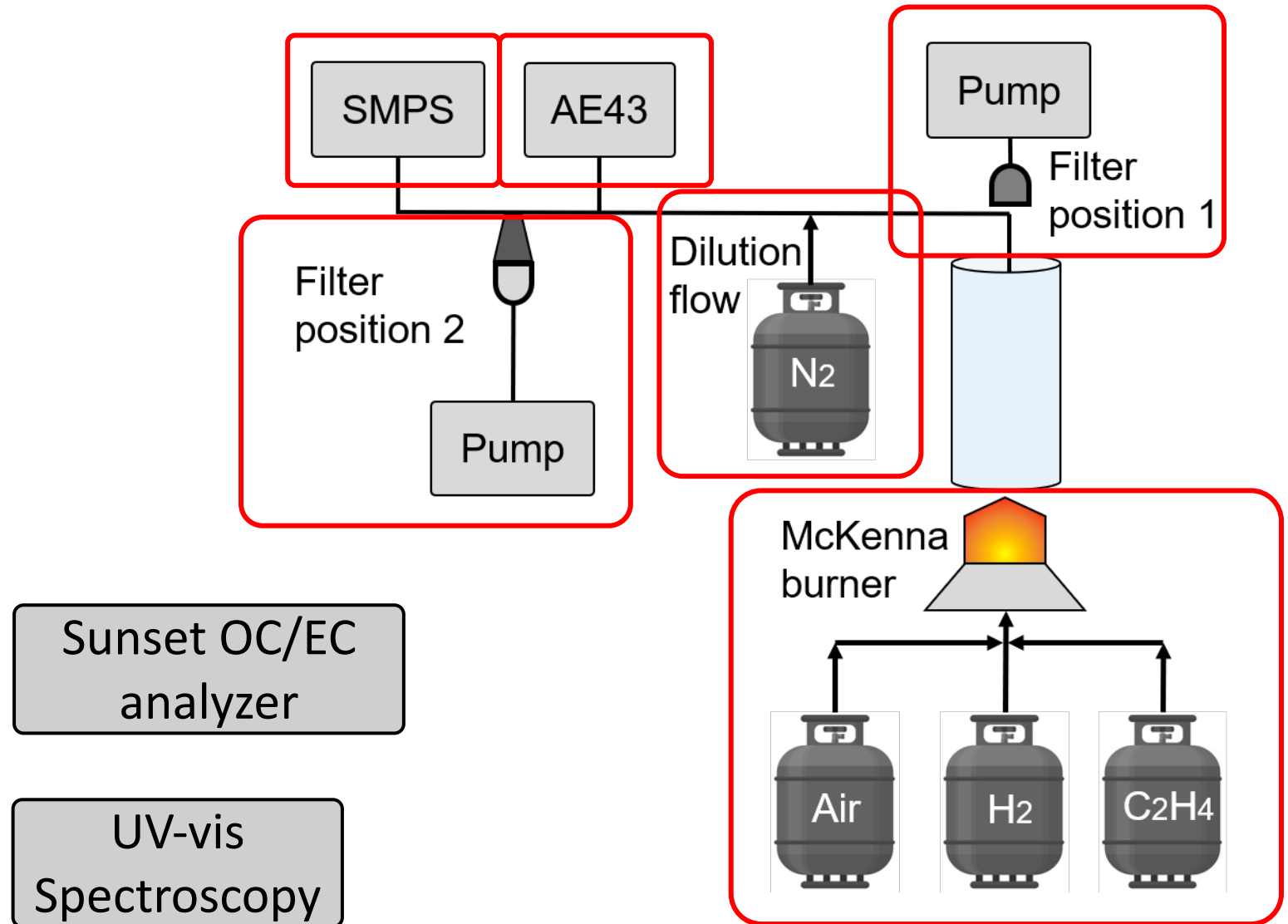
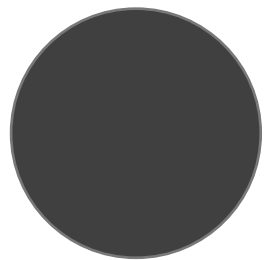
Flames of interest

- Fuel rich ($\phi = 2.3$)
- Premixed laminar flames
- Fuels: Ethylene (C_2H_4) and Hydrogen
- Hydrogen mole fraction α ranges from 0 – 0.5
- Fuel exit velocity ranging from 6 – 10 cm/s to parameterise flame temperature
- *Do the optical, physical and thermochemical properties of the aerosol change as a function of hydrogen addition and/or flame temperature?*



Experimental setup

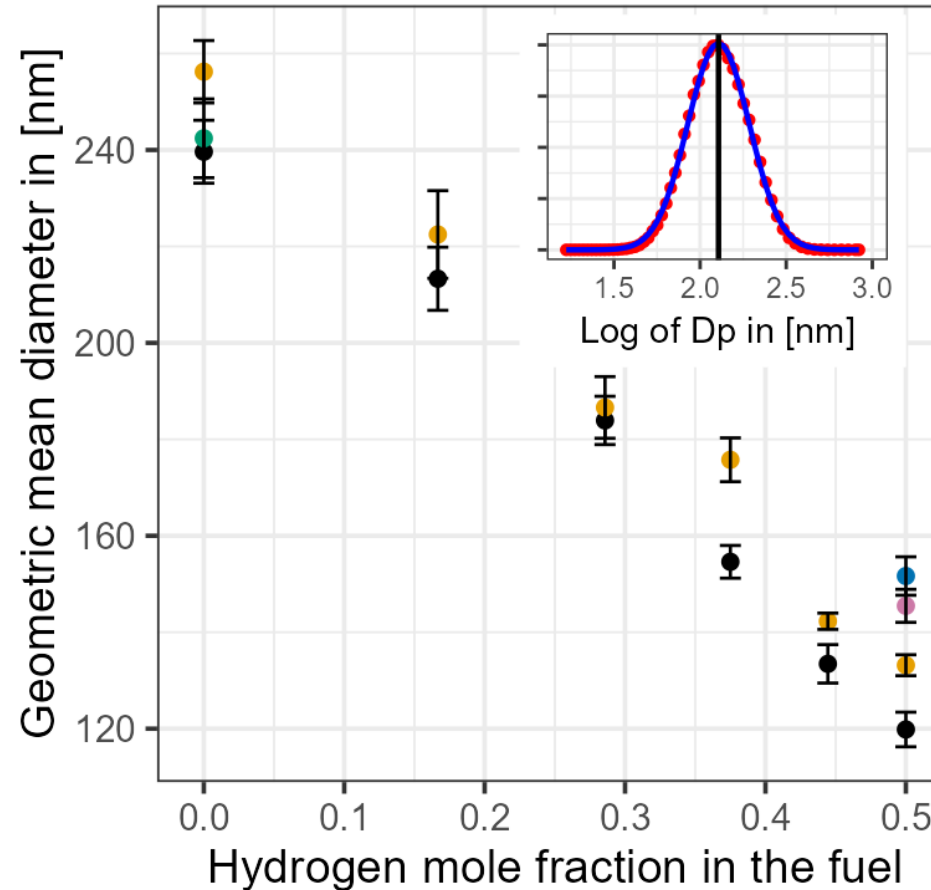
- Flame setup
- Dilution with Nitrogen
- Particle size distribution
- Optical absorption spectrum
- Filter sampler
- Mass of OC and EC
- Optical properties of OC



Results: Count particle distribution



- Lognormal distribution
- No bimodal distribution
- Smaller particles for high hydrogen content
- Exit velocity is less impactful



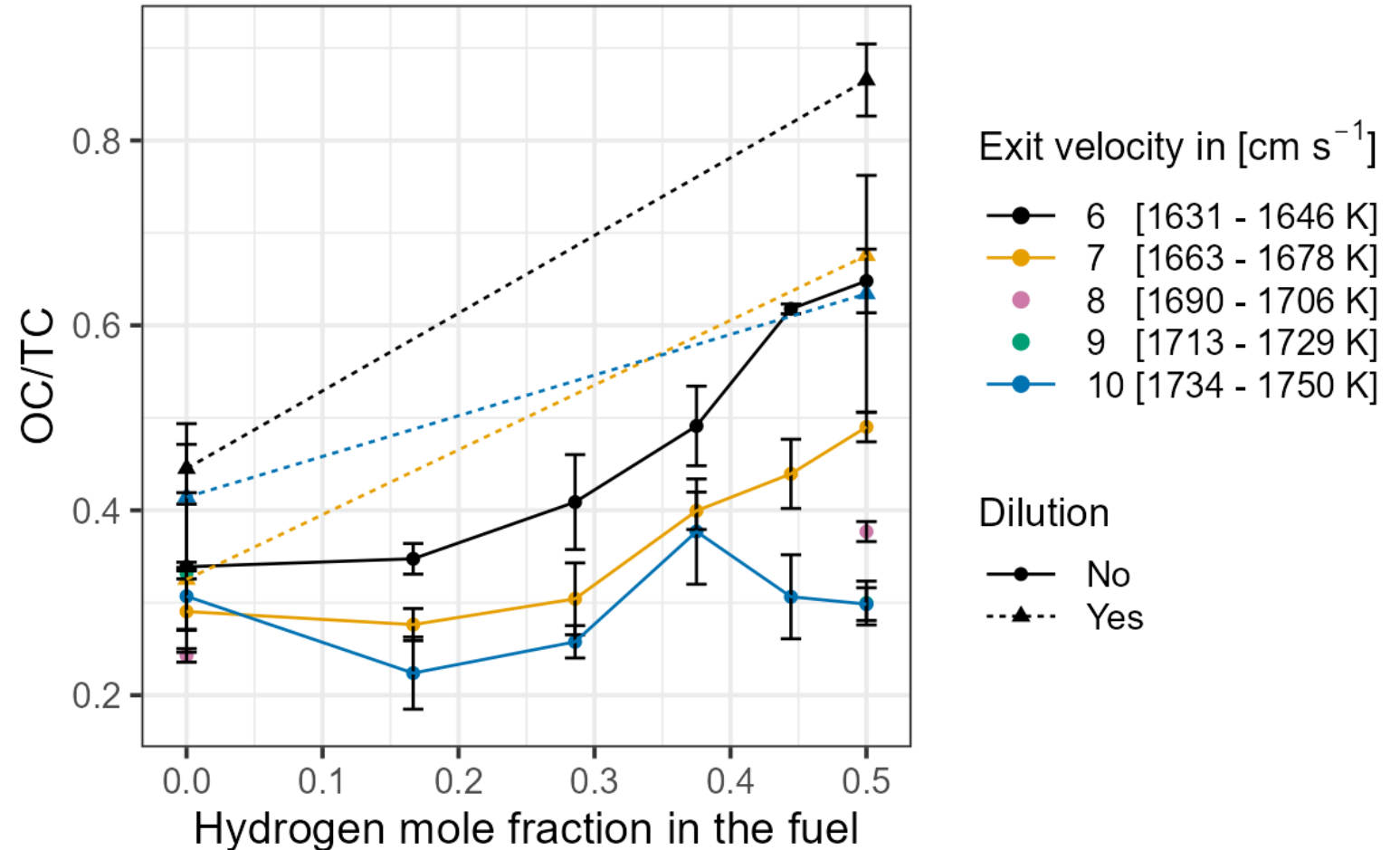
Exit velocity in [cm s⁻¹]

- 6 [1631 - 1646 K]
- 7 [1663 - 1678 K]
- 8 [1690 - 1706 K]
- 9 [1713 - 1729 K]
- 10 [1734 - 1750 K]



Results: Thermochemical properties

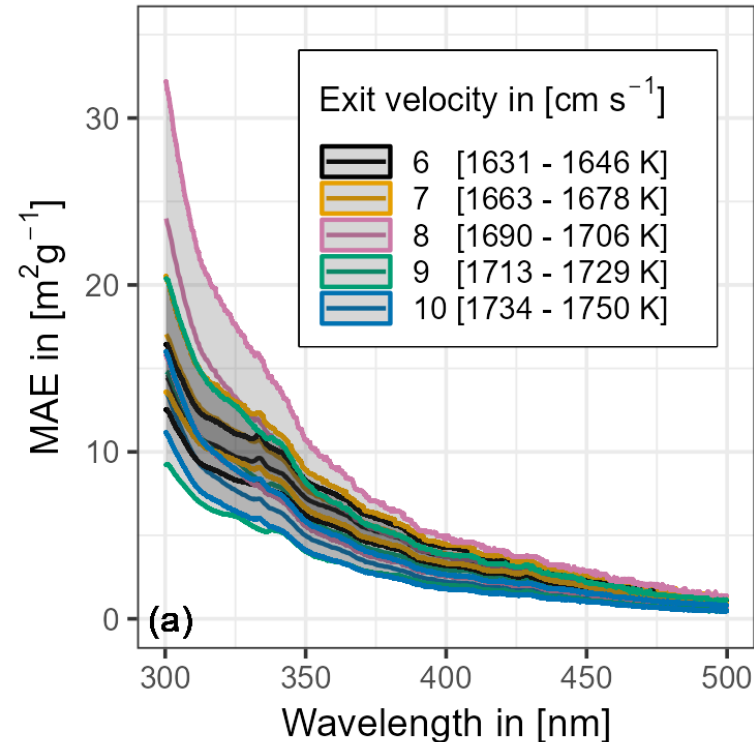
- Example of filters
- Sunset OC/EC analyser
- EC dominant aerosol for low hydrogen content and or high exit velocities
- For high H₂ content **combined** with low exit velocity, OC is dominant
- Dilution increases OC/TC





Results: Optical properties of OC

- Mass Absorption Efficiency
- Methanol dissolved, not PM
- No absorption after 500 nm
- Material doesn't change with flame parameters
- Angström Absorption Exponent
- Same conclusion



$$MAE_{OC,\lambda} = \frac{b_{abs,dis,\lambda}}{c_{OC}}$$

$$C_{abs}(\lambda) = C_{abs}(\lambda_0) * \left(\frac{\lambda}{\lambda_0}\right)^{-AAE}$$

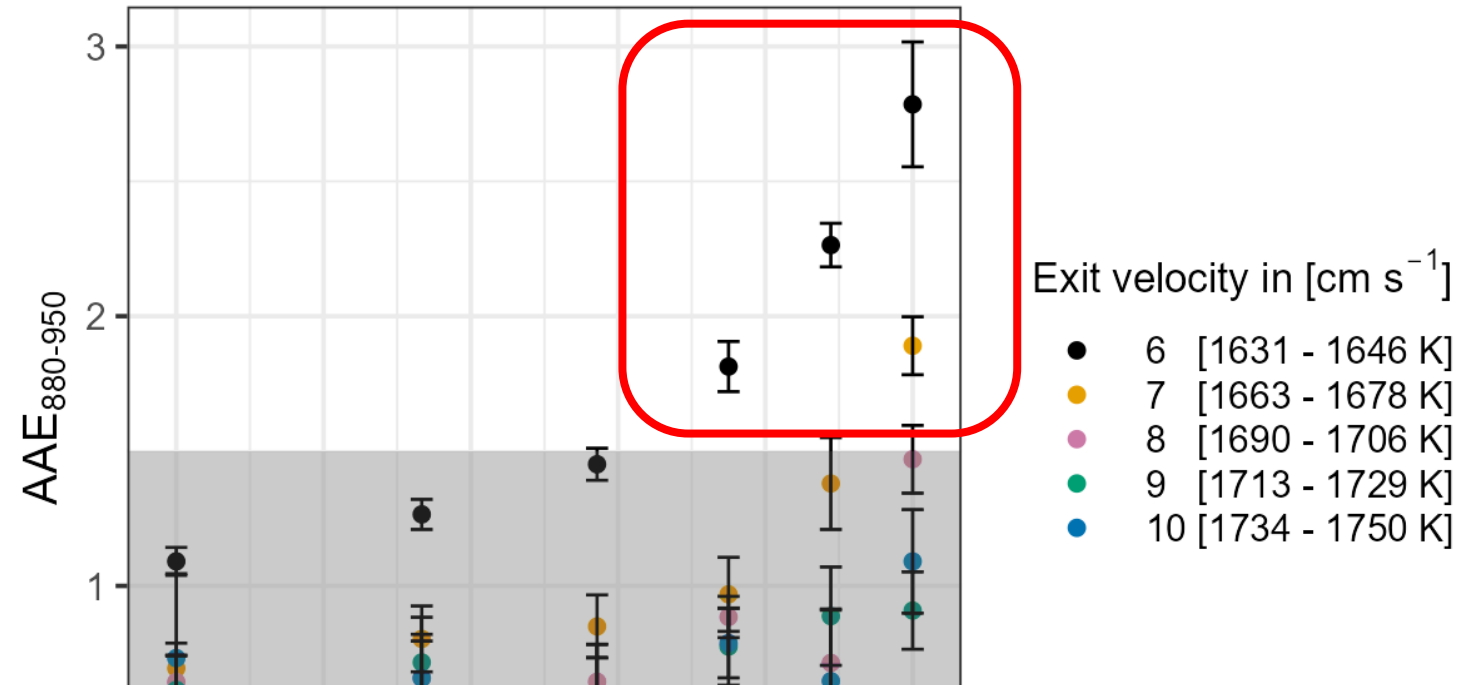
Results: Optical properties



EC

- AAE 880-950
- Grey area denoted common BC values
- BC seems possible, except for these flame conditions
- Compare with MAC_{EC}
- MAC_{EC} of pure BC is 4.68 m^2/g
- We conclude: Our EC does NOT resemble pure BC

$$MAC_{EC,\lambda=880} = \frac{b_{abs,\lambda=880}}{c_{EC}}$$

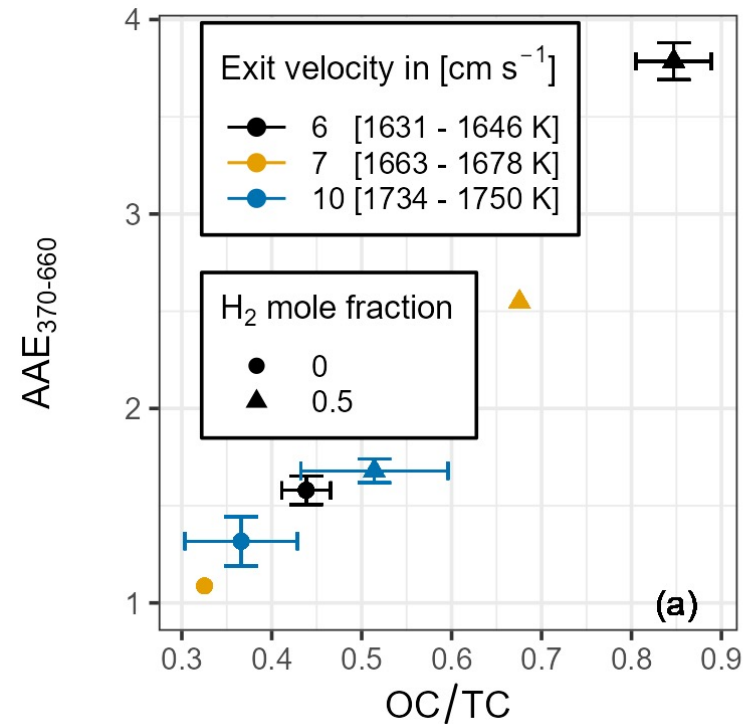


Exit velocity in [$cm s^{-1}$]	Hydrogen mole fraction	EC MAC 880 in [$m^2 g^{-1}$]
6	0	11.5 ± 0.5
6	0.5	5.9 ± 0.5
7	0	7.7
7	0.5	9.1
10	0	9.8 ± 0.8
10	0.5	11.8 ± 0.6



Discussion

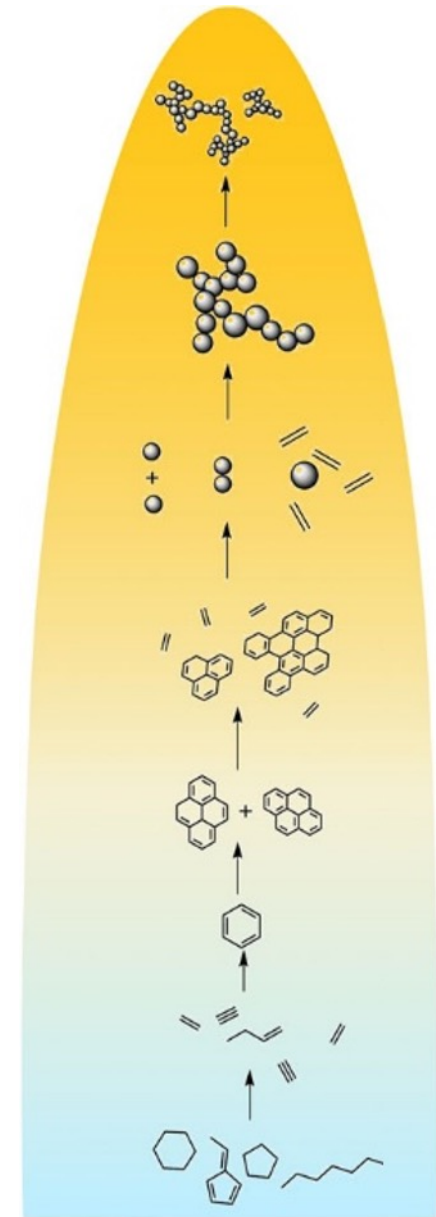
- Relate optical properties to OC/TC
- Very clear relation between AAE and OC/TC
- Relate MAC_{TC} to OC/TC
- Increased MAC_{TC} at first
- Later on, MAC_{TC} decreases
- Lensing effect could be an explanation





Discussion points

- Hydrogen addition decreases the particle size, but increases the volatility and absorption in shorter wavelengths
- Enhanced absorption in shorter wavelengths due to coloured OC
- Our EC does not resemble pure BC
- Part of the OC that we found is a precursor to the EC
- The lensing effect could explain some of the increased light absorption efficiency
- Questions ?



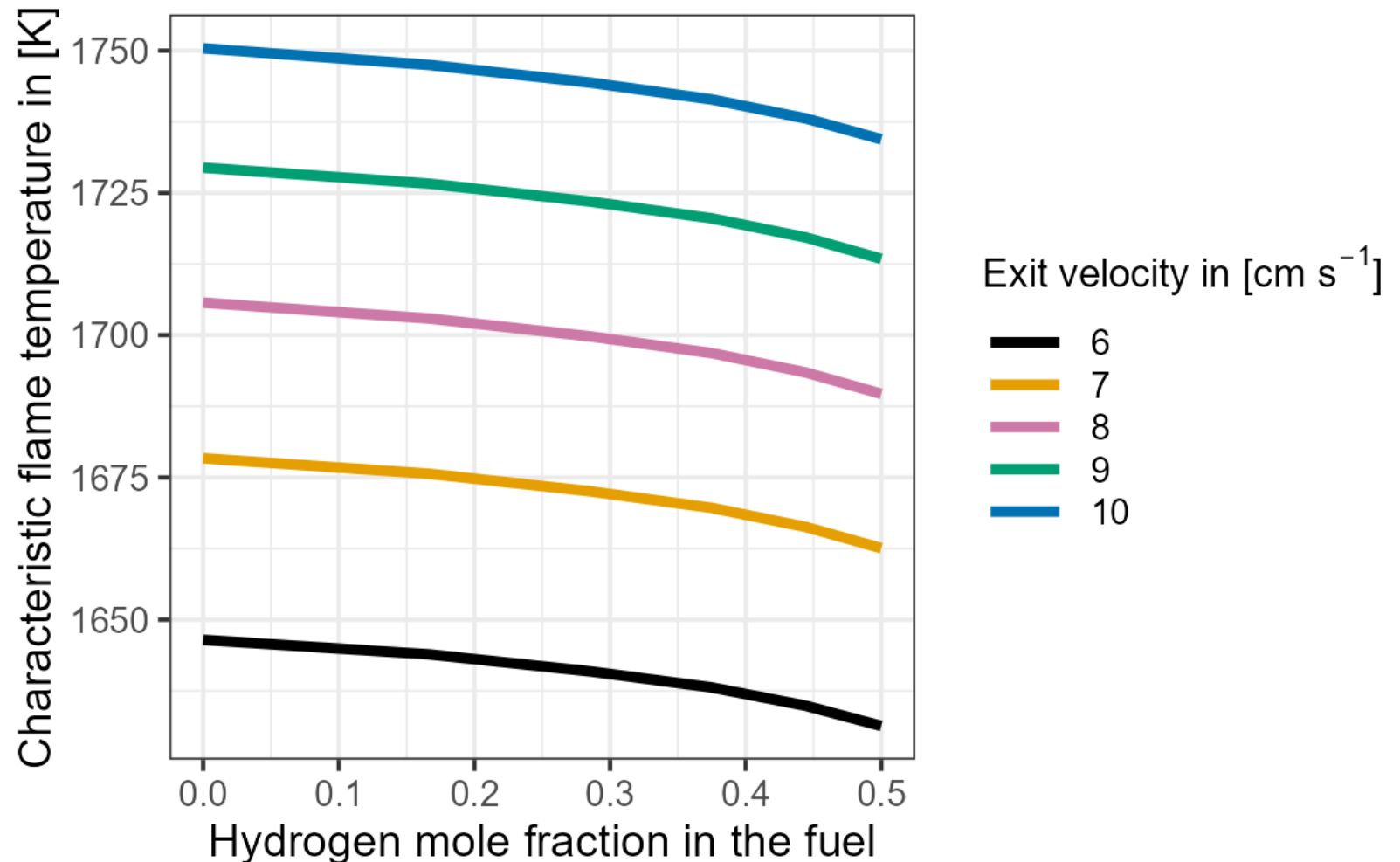
Backup slides





Flame temperature

- San Diego mechanism
- Characteristic temperature taken at HAB = 1
- Hydrogen addition cools the flame
- Exit velocity has larger impact

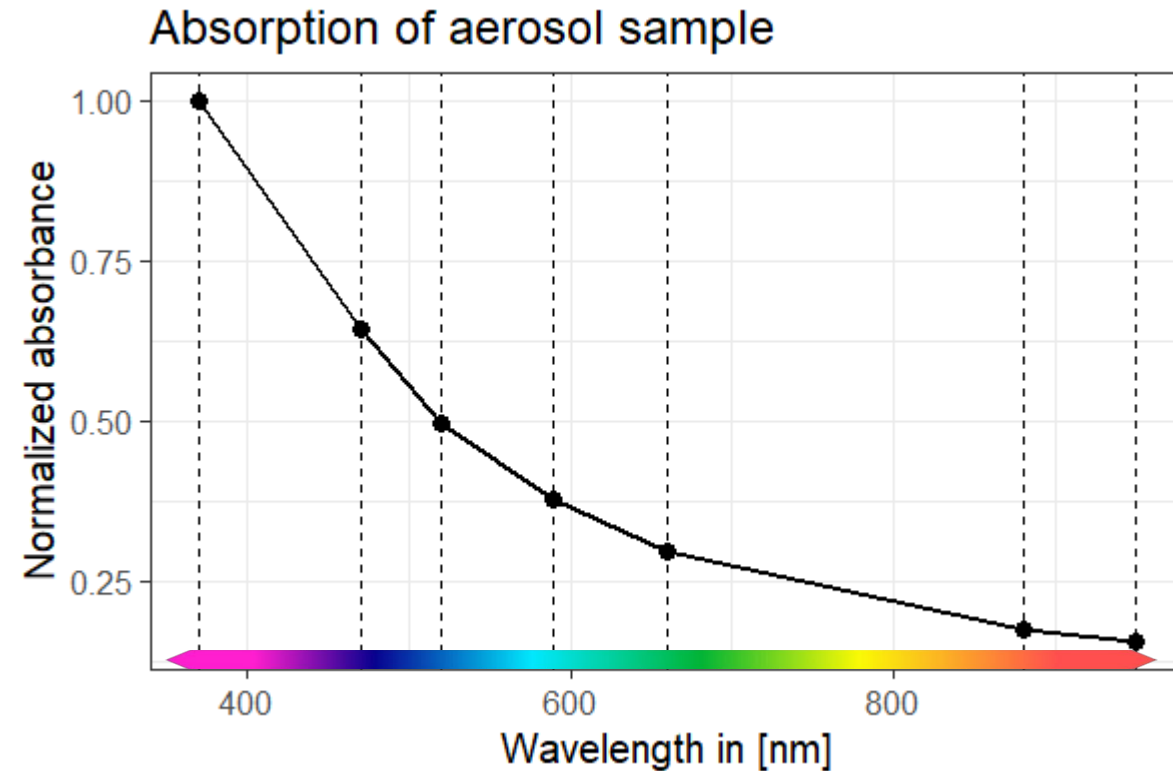




Optical properties

- Ex-situ -> Above the flame
- Wavelength depending absorption
- Absorption Angstrom exponent
- AAE of Black Carbon ≈ 1
- AAE of Brown Carbon > 1

$$C_{abs}(\lambda) = C_{abs}(\lambda_0) * \left(\frac{\lambda}{\lambda_0}\right)^{-AAE}$$



BC + BrC

BC + BrC

BC + BrC?

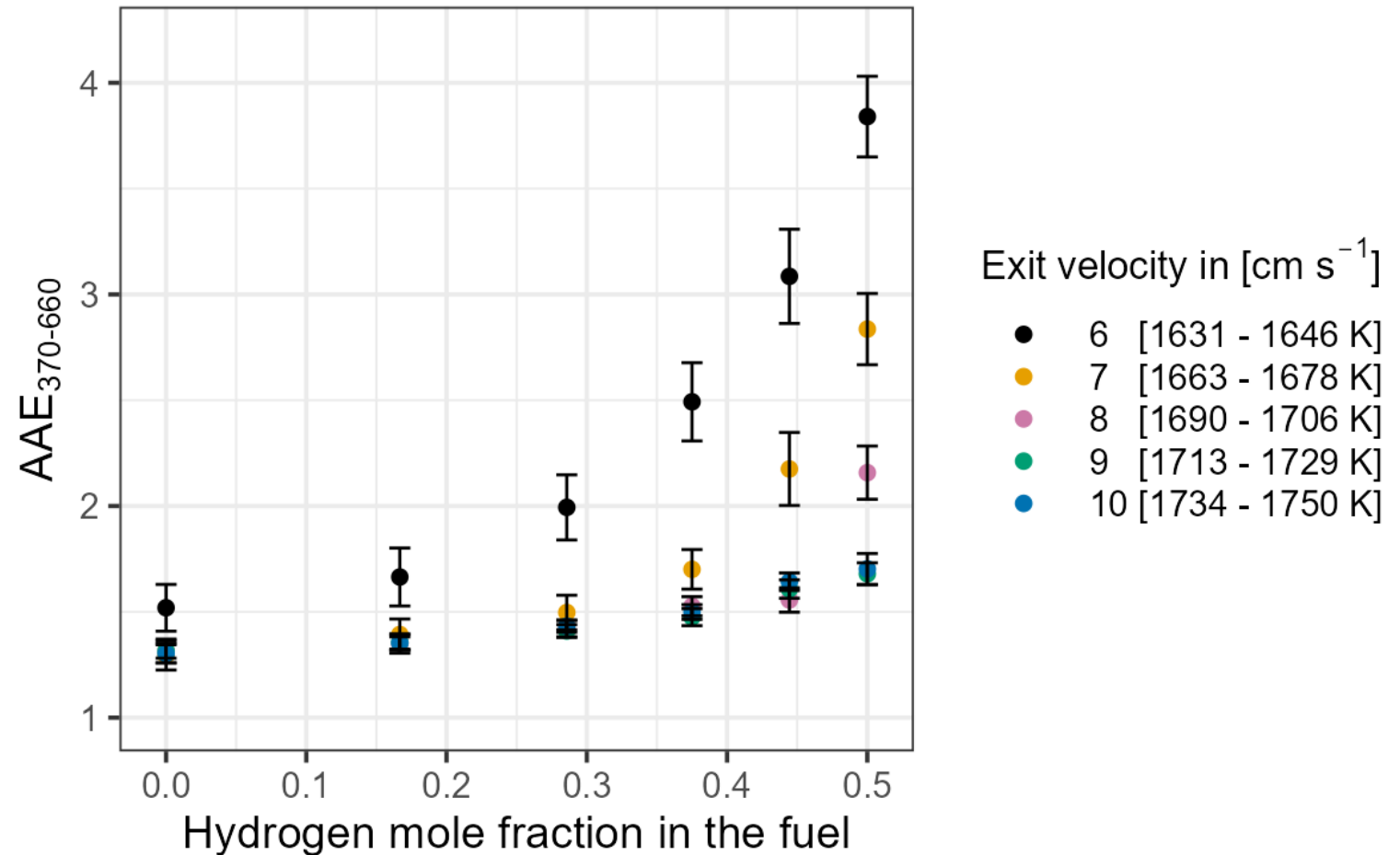
BC

BC



Optical properties TC

- AAE 370-660
- All above 1
- Resembles the OC/TC graph a lot
- For high hydrogen content **combined** with low exit velocity, OC is dominant



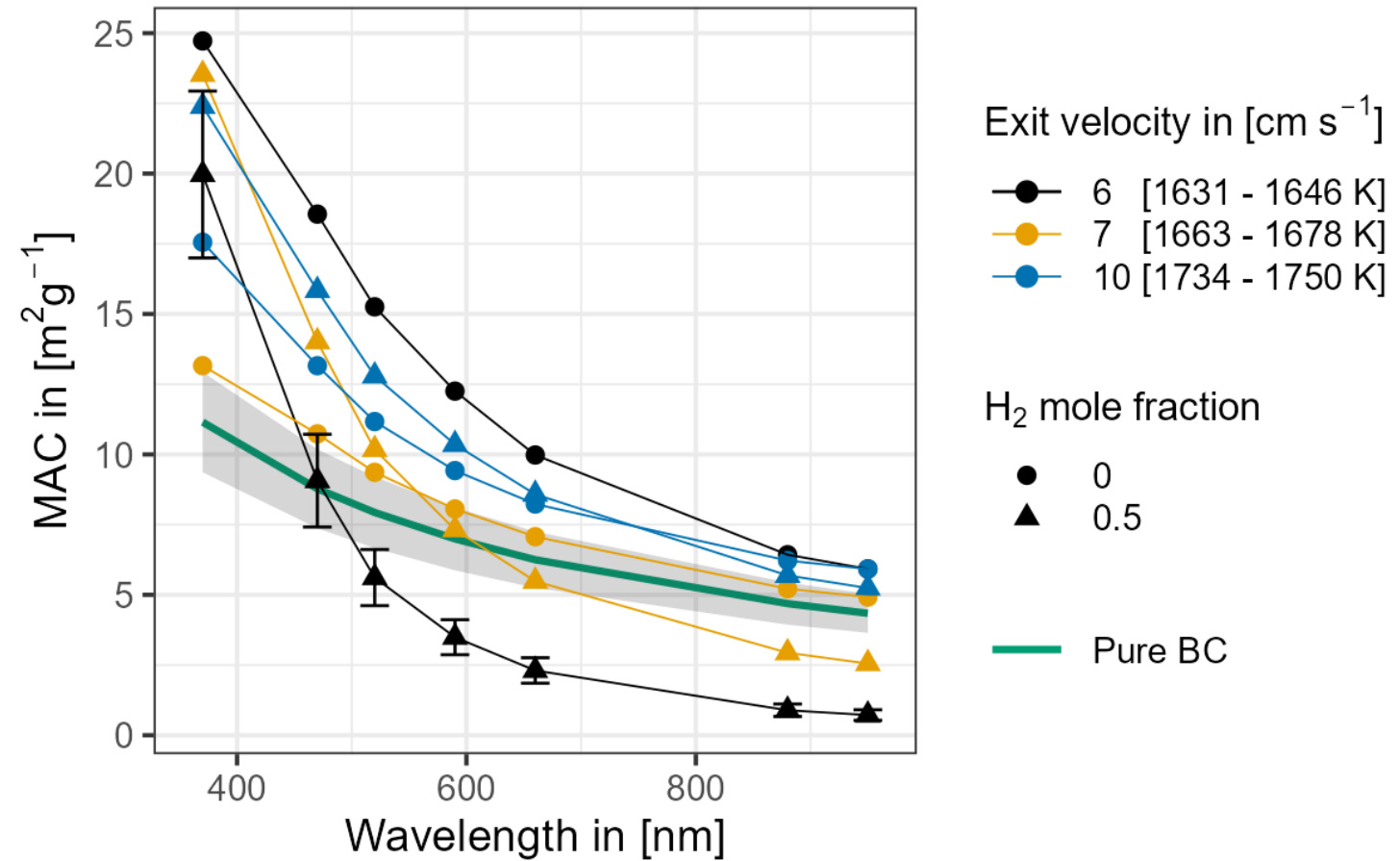
Results: Optical properties



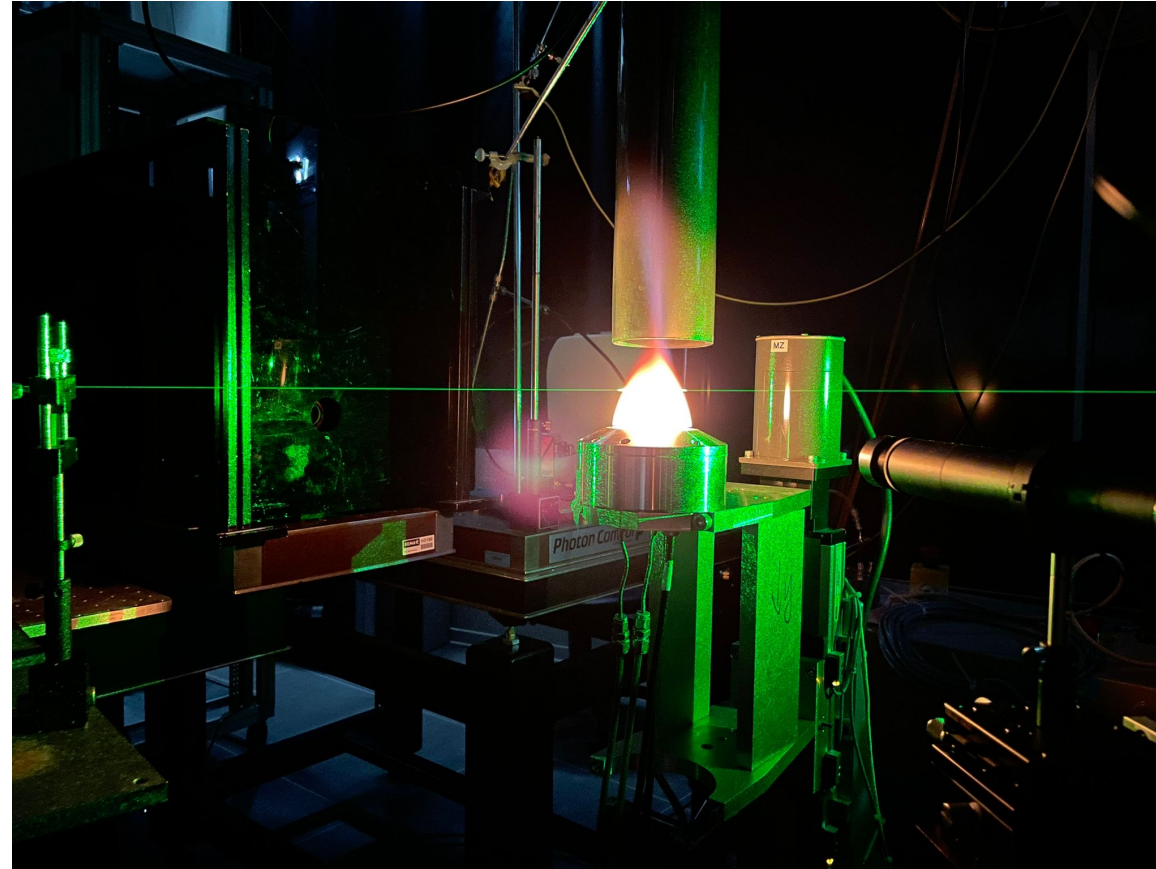
TC

- TC Mass Absorption Cross section
- Green line is pure BC
- OC increase absorption in the shorter wavelengths.
- Longer wavelengths seem ambiguous

$$MAC_{TC,\lambda} = \frac{b_{abs,\lambda}}{c_{TC}}$$

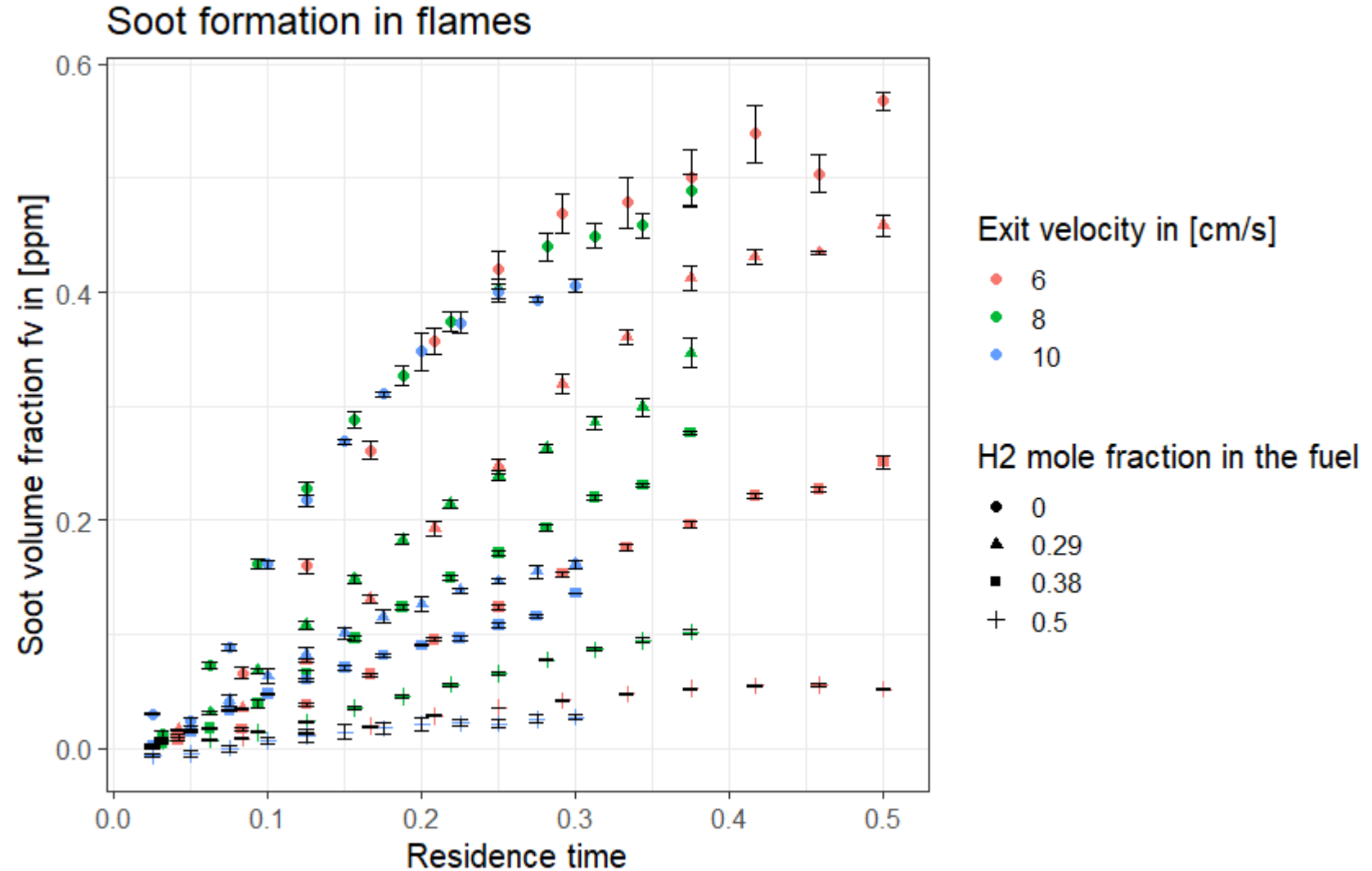
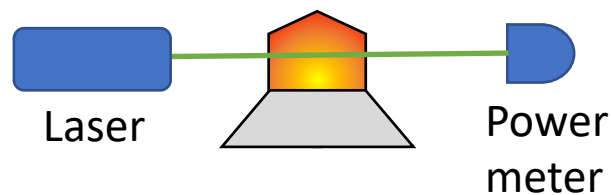


BC volume fraction by LLE



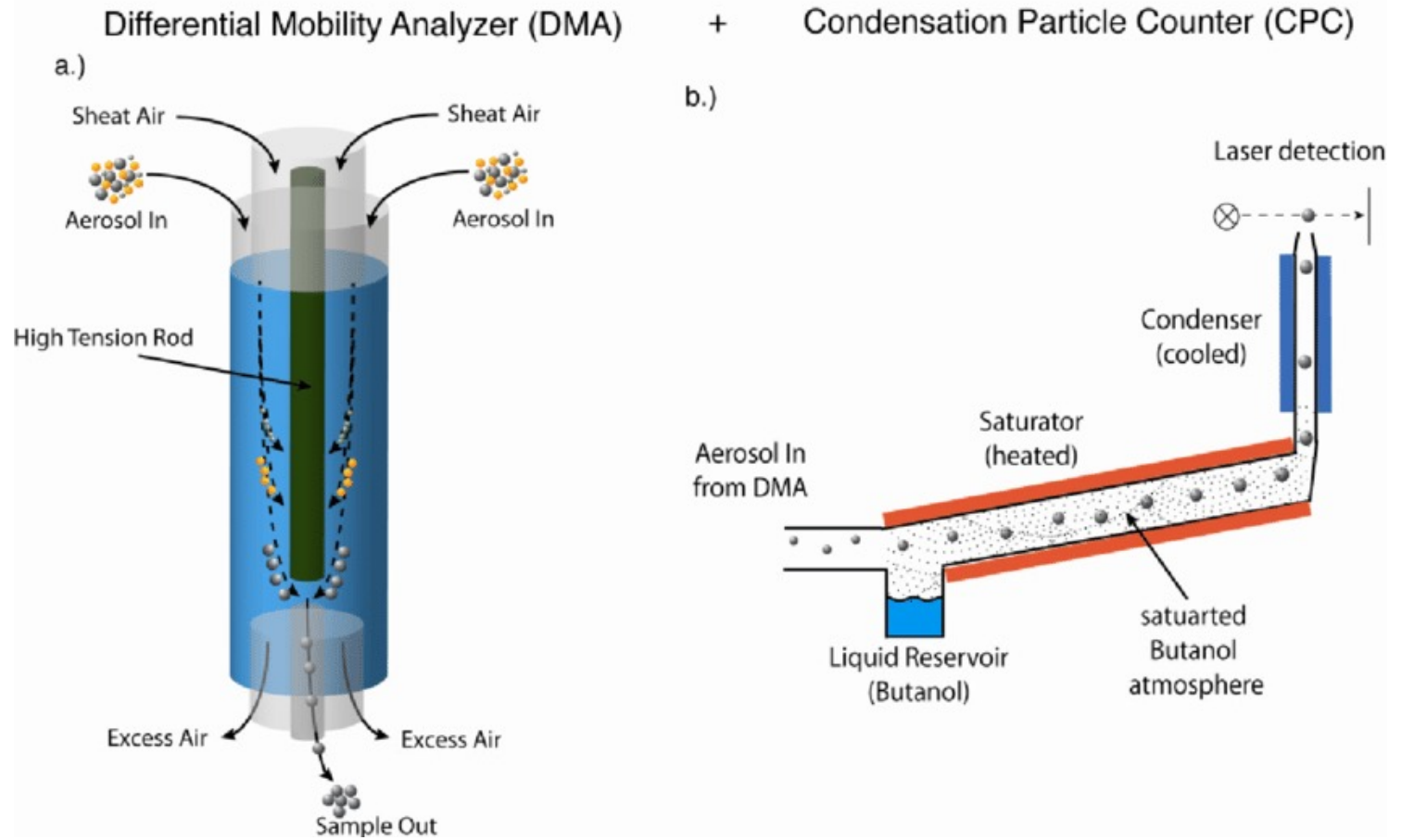
BC volume fraction by LLE

- Laser light extinction
- Measure decrease in laser intensity
- Differentiate HAB, calculate residence time
- BC volume fraction reduction by H₂ addition



Working principle SMPS

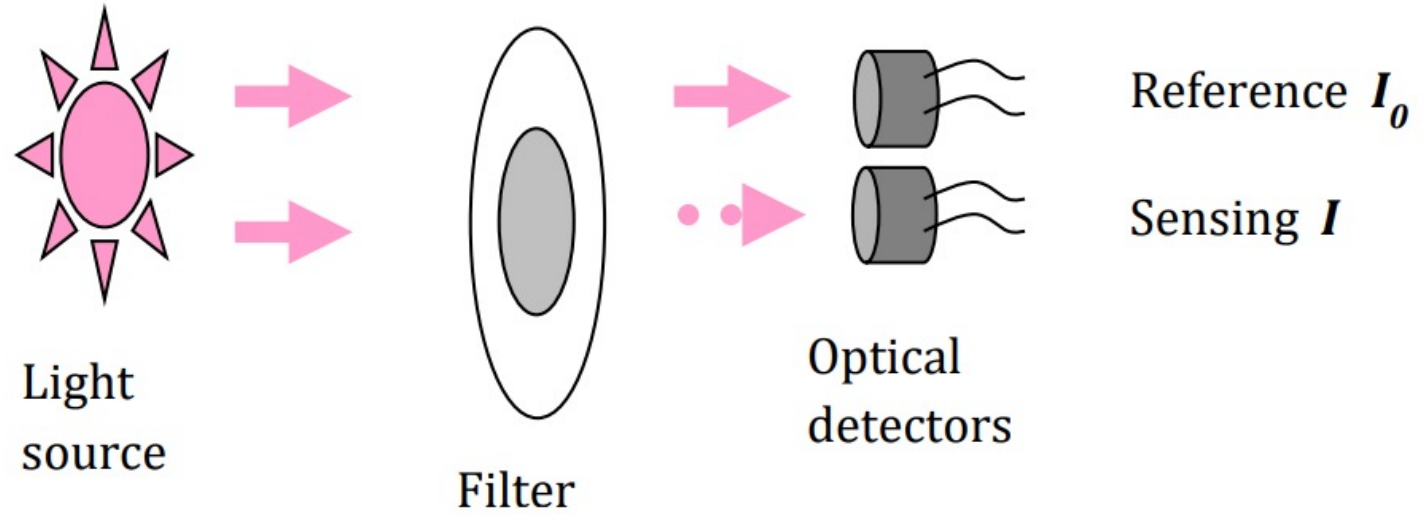
- Particle sizer
- DMA
- CPC



Working principle Aethalometer



- Wavelength depending absorption
- Sample deposition
- Dual spot
- Attenuation on 7 wavelengths



Working principle Sunset OC/EC



- Sample filter
- Heat in helium environment for OC extraction
- Heat in helium + oxygen environment for EC extraction
- Measure CO₂ from extracted carbon

