

STUDY OF SOOT PRODUCTION IN ETHYLENE PYROLYSIS USING A SECTIONAL MODEL

24th ETH-Conference on Combustion Generated Nanoparticles
June 22-24, 2021 – Online Conference



Leonardo PACHANO

Damien AUBAGNAC-KARKAR

OUTLINE

- Background
- Objective and target conditions
- Soot modeling approach
- Results
- Summary

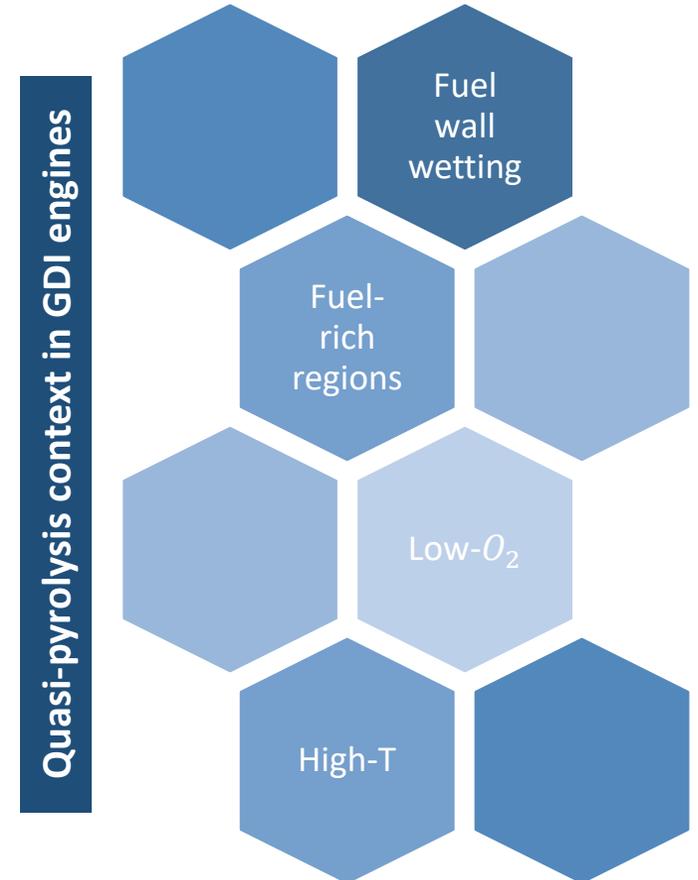
BACKGROUND

Nanoparticle formation from hydrocarbons oxidation and pyrolysis

- Of interest for the development of more efficient processes for nanoparticle synthesis
- Crucial for suppressing undesirable soot emissions that might impact the environment and human health



Video courtesy of Kamal Shway Shway
kamal.shway-shway@ifpen.fr



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OBJECTIVE AND TARGET CONDITIONS

Modeling soot production in ethylene pyrolysis using a sectional approach

Long-term objective: Assessment in CFD pyrolyzing spray cases

- **PFR:** Soot concentration, PAH, ethylene and acetylene mole fraction at long residence times
- **Shock tube:** Time-resolved soot volume fraction and acetylene mole fraction

	PFR [1]	ST [2]
Experiment type	Plug flow reactor	Shock tube
X_{fuel}	0.03	0.0235
Diluent	Nitrogen	Argon
T [K]	1223-1423	2179
P [Pa]	1.01×10^5	3.80×10^5
Time [ms]	$\sim 10^3$	$\sim 10^1$

[1] Sánchez, Nazly E. Estudio de la formación de hidrocarburos aromáticos policíclicos (hap) en la pirólisis de acetileno y etileno. Diss. Universidad de Zaragoza, 2014.

[2] Utsav, K. C., Mohamed Beshir, and Aamir Farooq. "Simultaneous measurements of acetylene and soot during the pyrolysis of ethylene and benzene in a shock tube." Proceedings of the Combustion Institute 36.1 (2017): 833-840.

SOOT MODELING APPROACH

Pachano Prieto, L. M. (2020). CFD modeling of combustion and soot production in Diesel sprays (Doctoral dissertation).

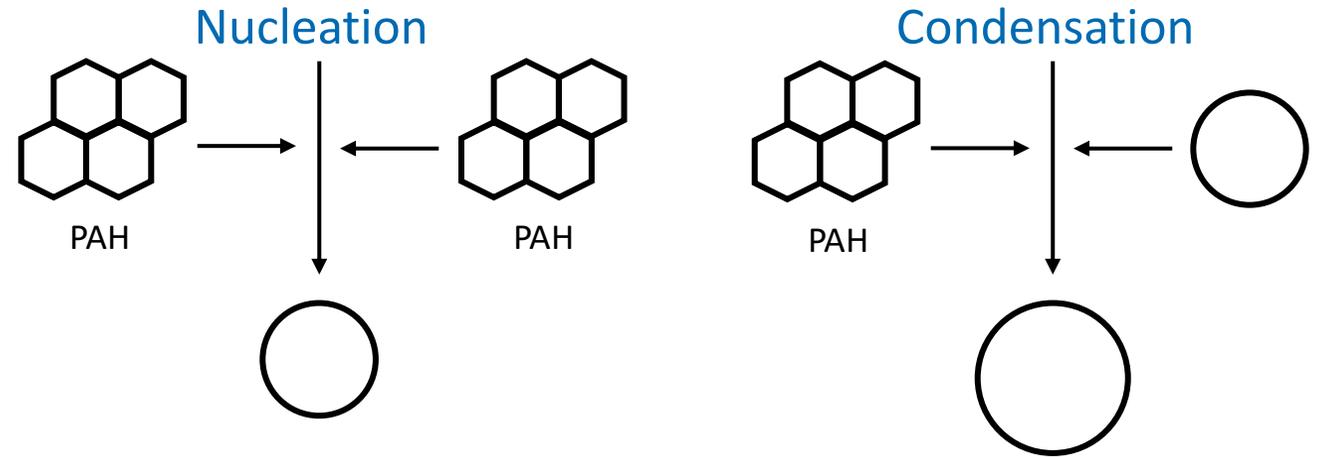
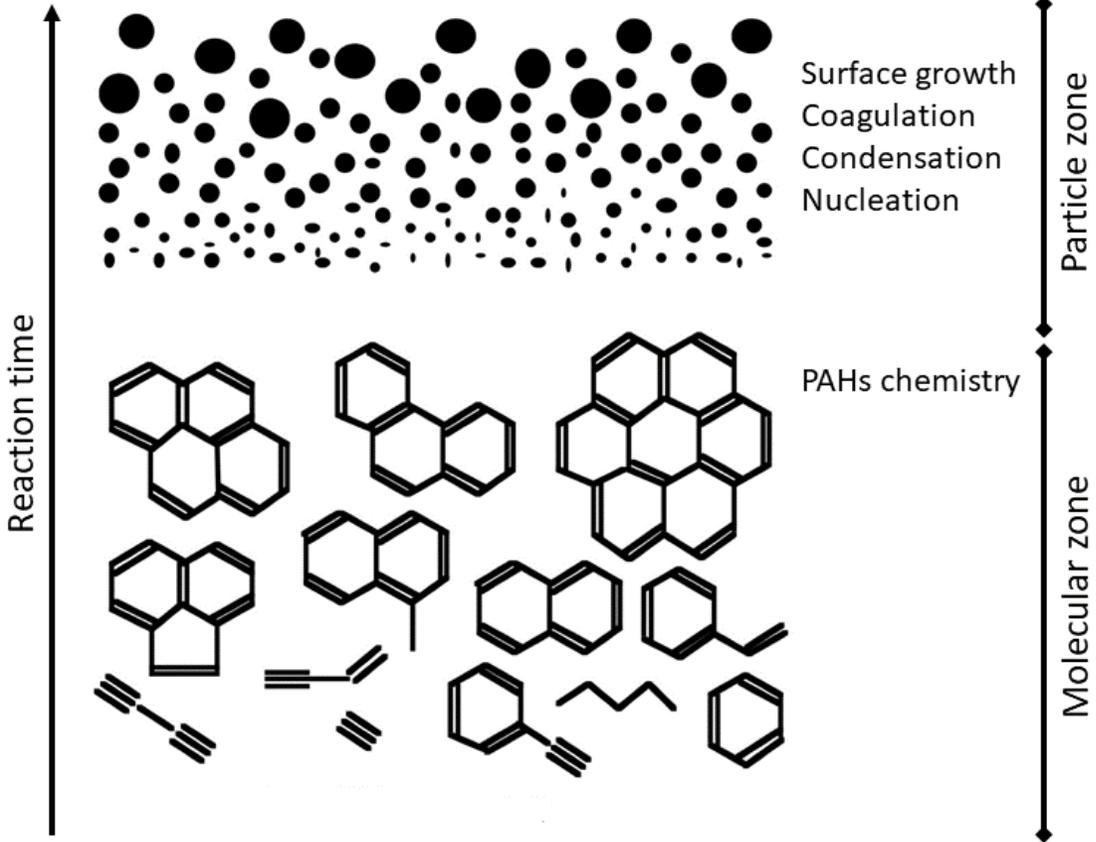
Adapted from: J. Warnatz, U. Maas, and R. W. Dibble. Combustion. Vol. 4. Berlin: Springer, 2006



Zero-dimensional simulations using Cantera and a Sectional Soot Model (SSM) [1,2]

Modeled collisional phenomena and surface chemistry

Processes increasing soot mass



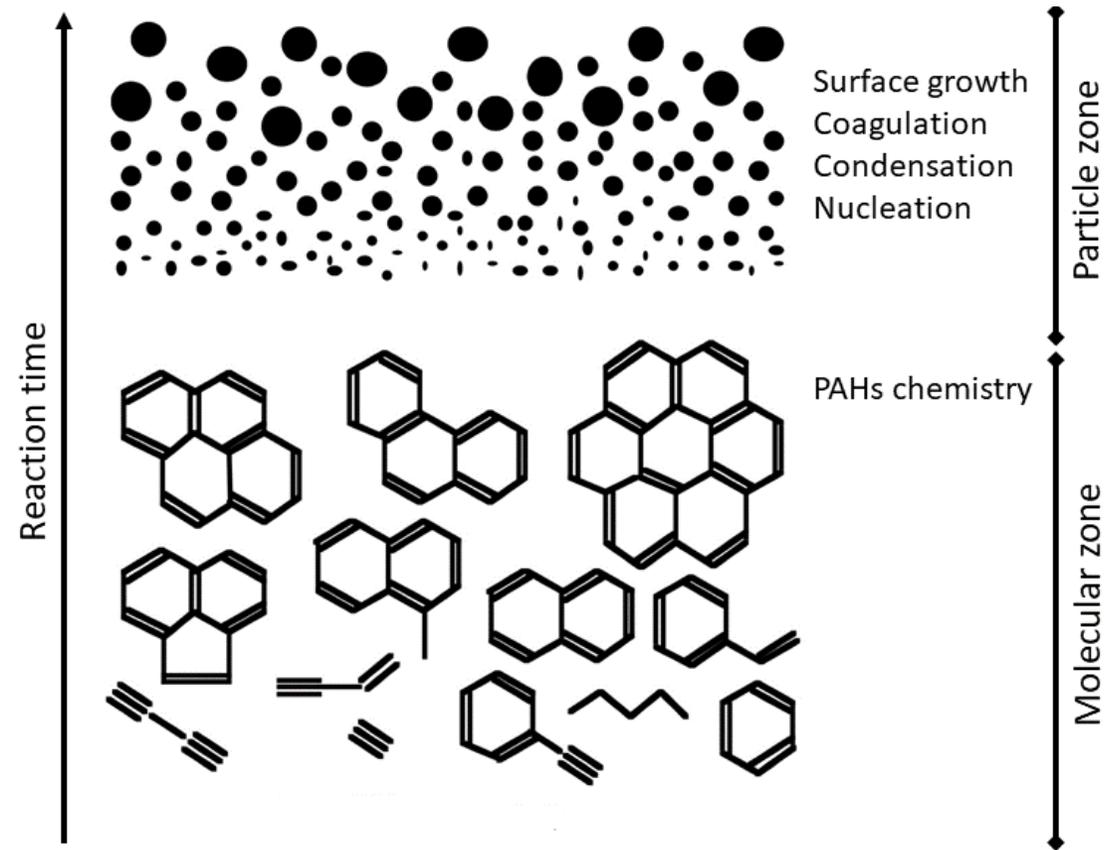
Reference numerical setup	
Gas phase coupling	Two-way
Nucleation model	Reversible with dampening factor
Condensation model	Reversible
Solver	Cantera 2.5.0a3

[1] Aubagnac-Karkar, D., Michel, J. B., Colin, O., Vervisch-Kljakic, P. E., & Darabiha, N. (2015). Sectional soot model coupled to tabulated chemistry for Diesel RANS simulations. Combustion and Flame, 162(8), 3081-3099.
 [2] Aubagnac-Karkar, D., El Bakali, A., & Desgroux, P. (2018). Soot particles inception and PAH condensation modelling applied in a soot model utilizing a sectional method. Combustion and Flame, 189, 190-206.

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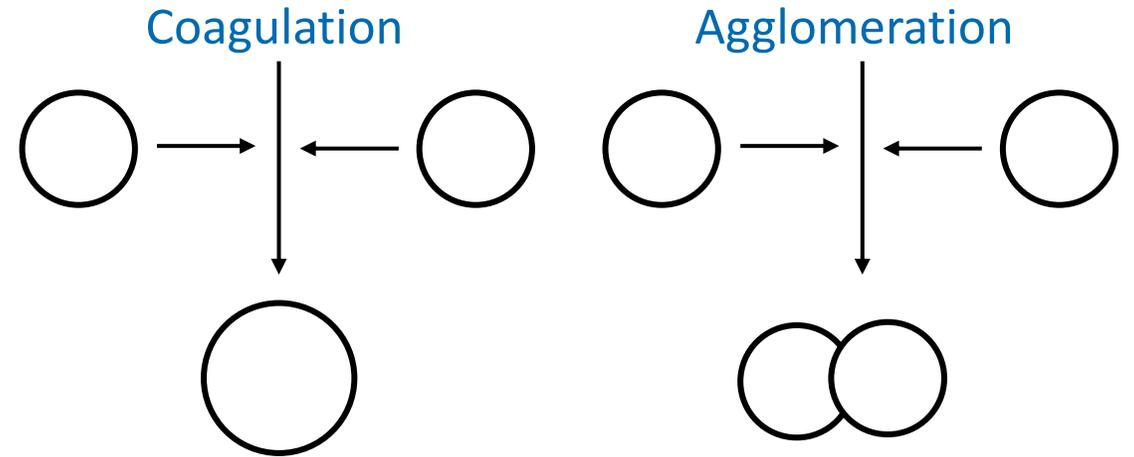
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Zero-dimensional simulations using Cantera and a Sectional Soot Model (SSM) [1,2]

Modeled collisional phenomena and surface chemistry

Processes acting on the number of particles



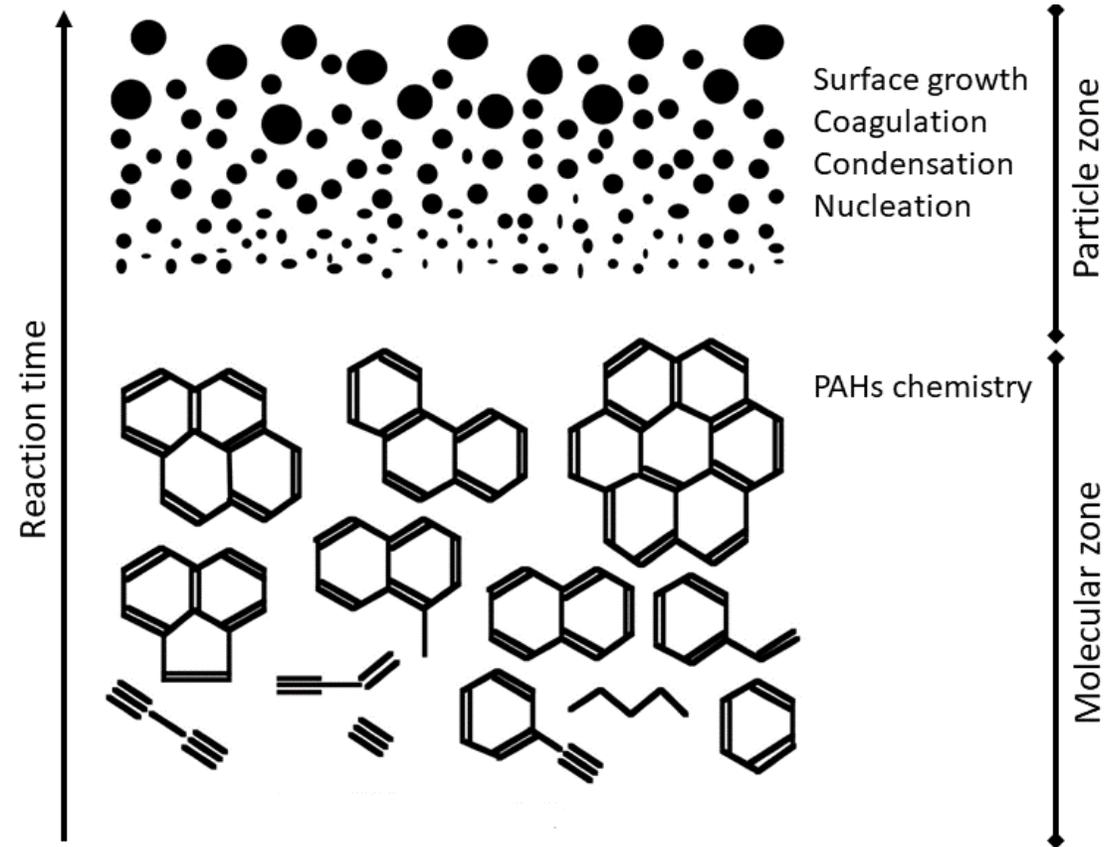
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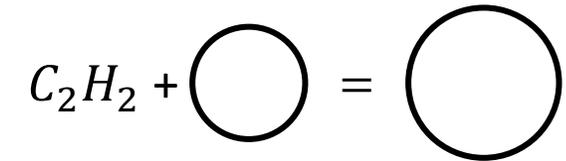


Zero-dimensional simulations using Cantera and a Sectional Soot Model (SSM) [1,2]

Modeled collisional phenomena and surface chemistry

Processes increasing soot mass

Surface growth

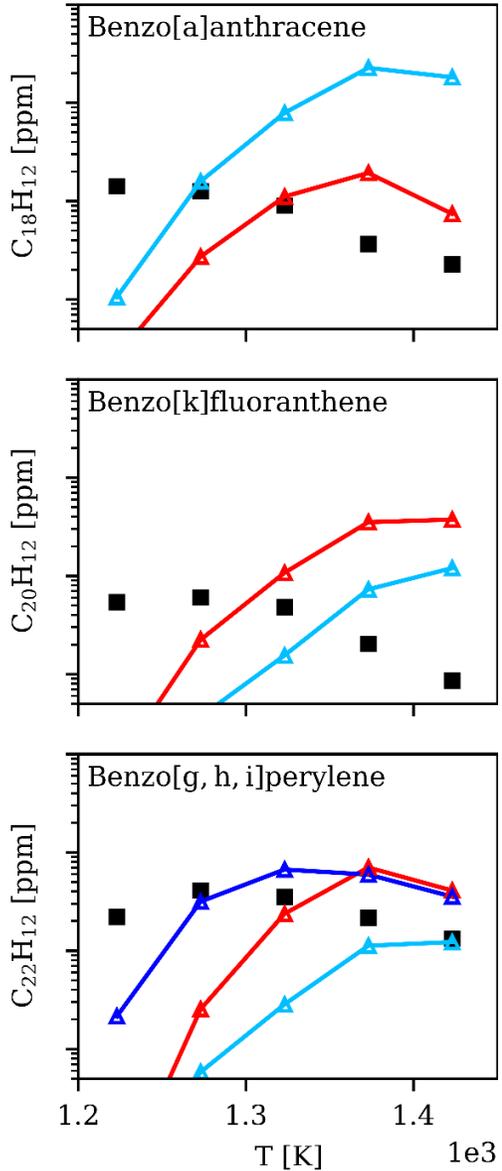
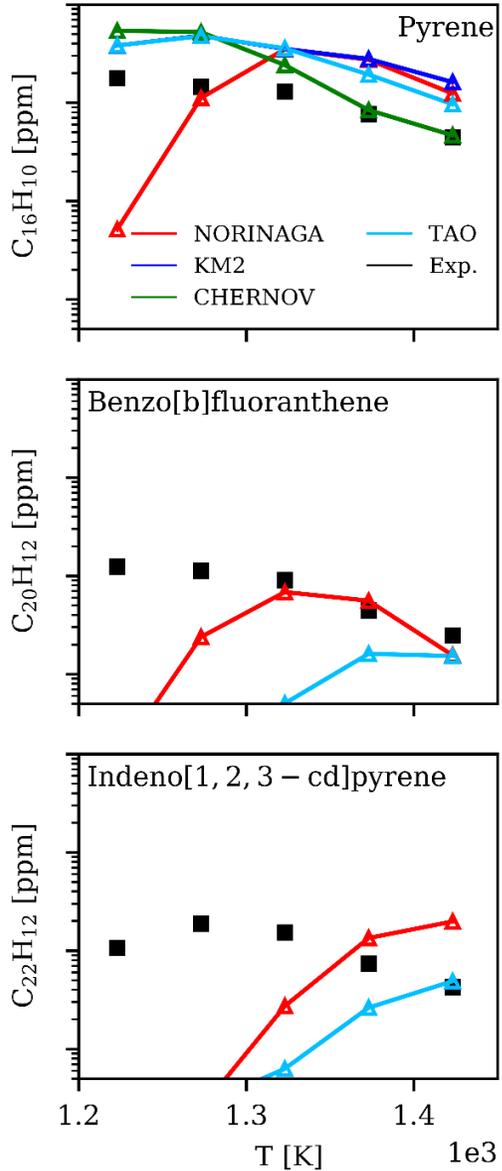
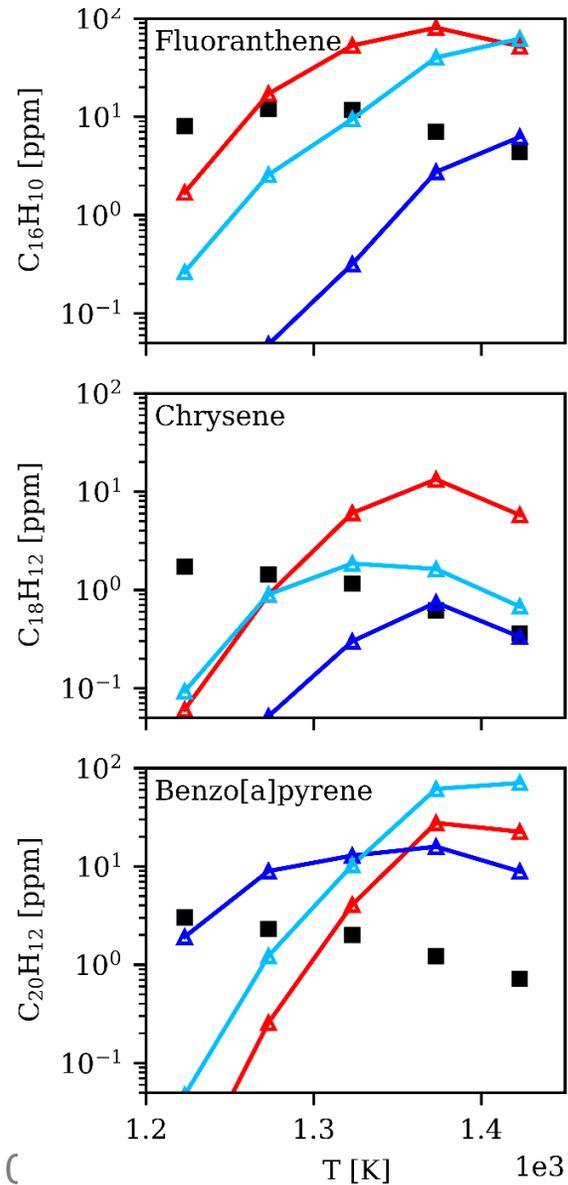


Reference numerical setup	
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PFR RESULTS



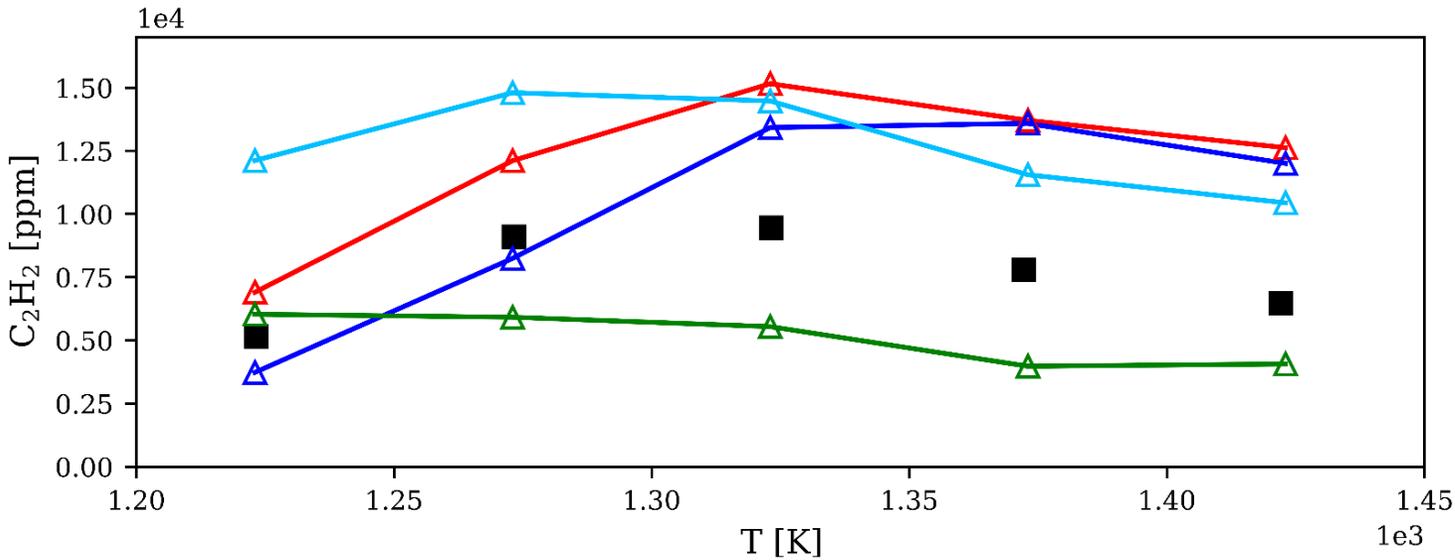
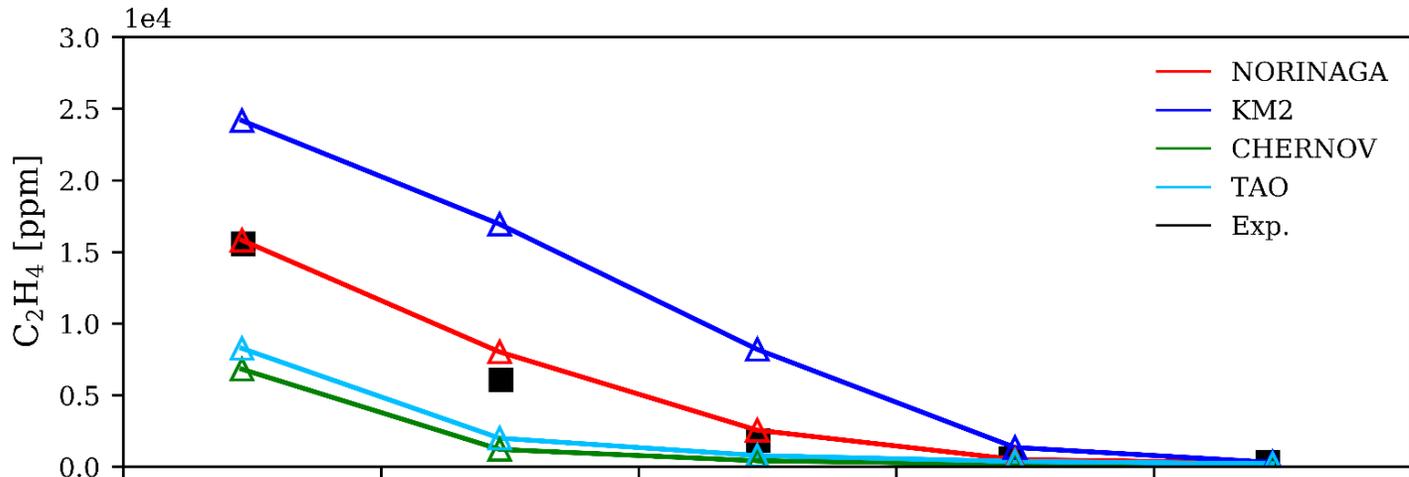
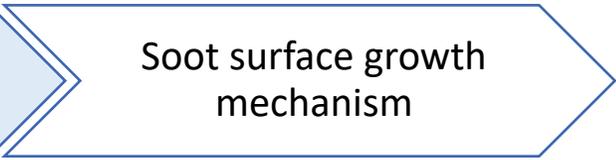
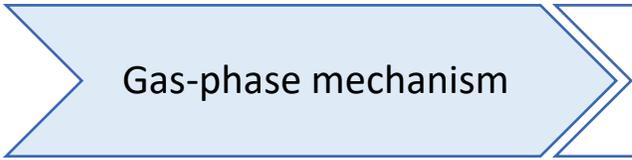
- PAH mole fraction results without soot modeling
- Choice of gas-phase mechanism inconclusive based on PAH results

Mechanism	Ref.
NORINAGA	[1]
KM2	[2]
CHERNOV	[3]
TAO	[4]

[1] Norinaga, Koyo, and Olaf Deutschmann. Industrial & engineering chemistry research 46.11 (2007): 3547-3557.
 [2] Wang, Yu, Abhijeet Raj, and Suk Ho Chung. Combustion and flame 160.9 (2013): 1667-1676.
 [3] Chernov, Victor, et al. Combustion and Flame 161.2 (2014): 592-601.
 [4] Tao, Hairong, et al. Fuel 255 (2019): 115796.

Experimental results from:
 Sánchez, Nazly E. Estudio de la formación de hidrocarburos aromáticos policíclicos (hap) en la pirólisis de acetileno y etileno. Diss. Universidad de Zaragoza, 2014.

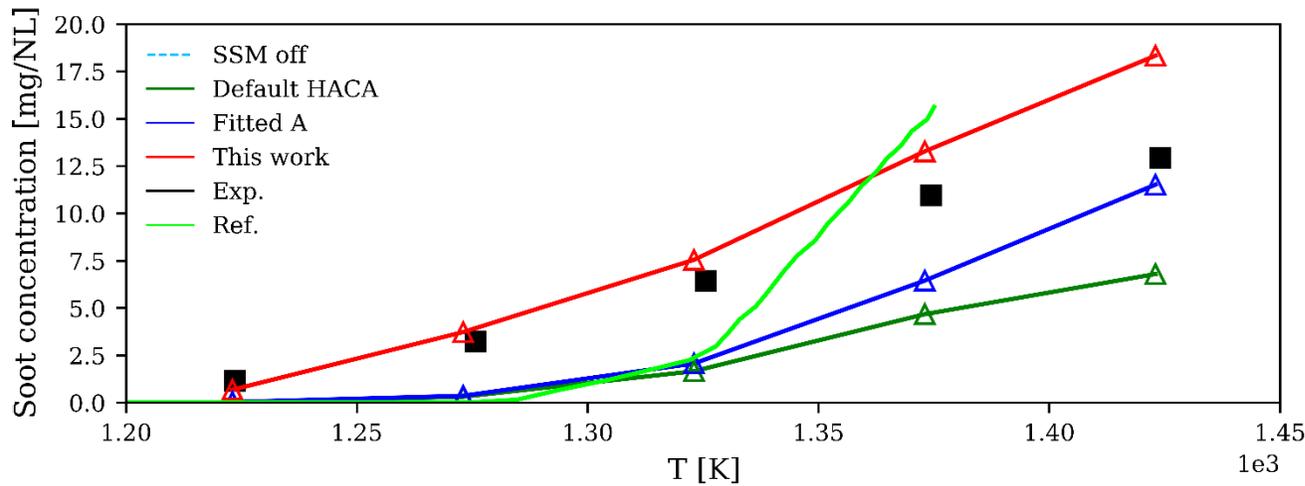
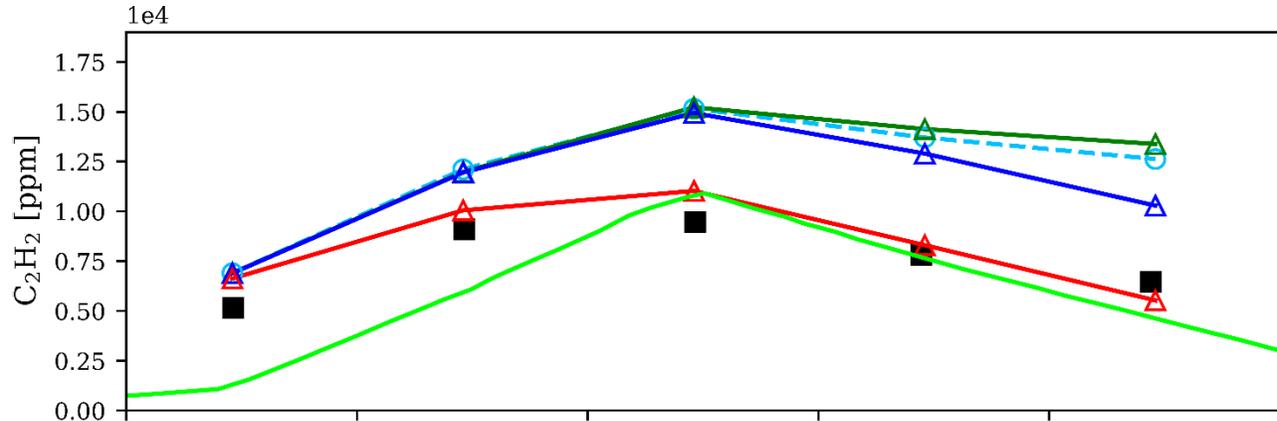
PFR RESULTS



- Fuel and acetylene mole fraction results
- NORINAGA is the best compromise despite differences in PAH results

Experimental results from:
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PFR RESULTS



- Reference results without soot modeling
 - C_2H_2 overprediction in line with no soot modeling
- Setup 1: Starting point
 - Pyrene and all bigger PAHs used as precursors
 - Surface growth follows HACA mechanism [1]
- Setup 2 : Fitting C_2H_2 addition pre-exponential factor (A)
- Setup 3: Enhanced surface growth through C_2H , CH_3 and C_3H_3 [2]

↓

	Setup 1	Setup 2	Setup 3
A	80	2500	80
Surface growth mechanism	$C_{soot} + H \rightleftharpoons C_{soot}^* + H_2$ $C_{soot}^* + H \rightarrow C_{soot}$ $C_{soot}^* + C_2H_2 \rightarrow C_{soot} + H$	HACA + $C_{soot} + C_2H \rightarrow C_{soot}^* + C_2H_2$ $C_{soot} + CH_3 \rightarrow C_{soot}^* + CH_4$ $C_{soot} + C_3H_3 \rightarrow C_{soot}^* + AC_3H_4$	

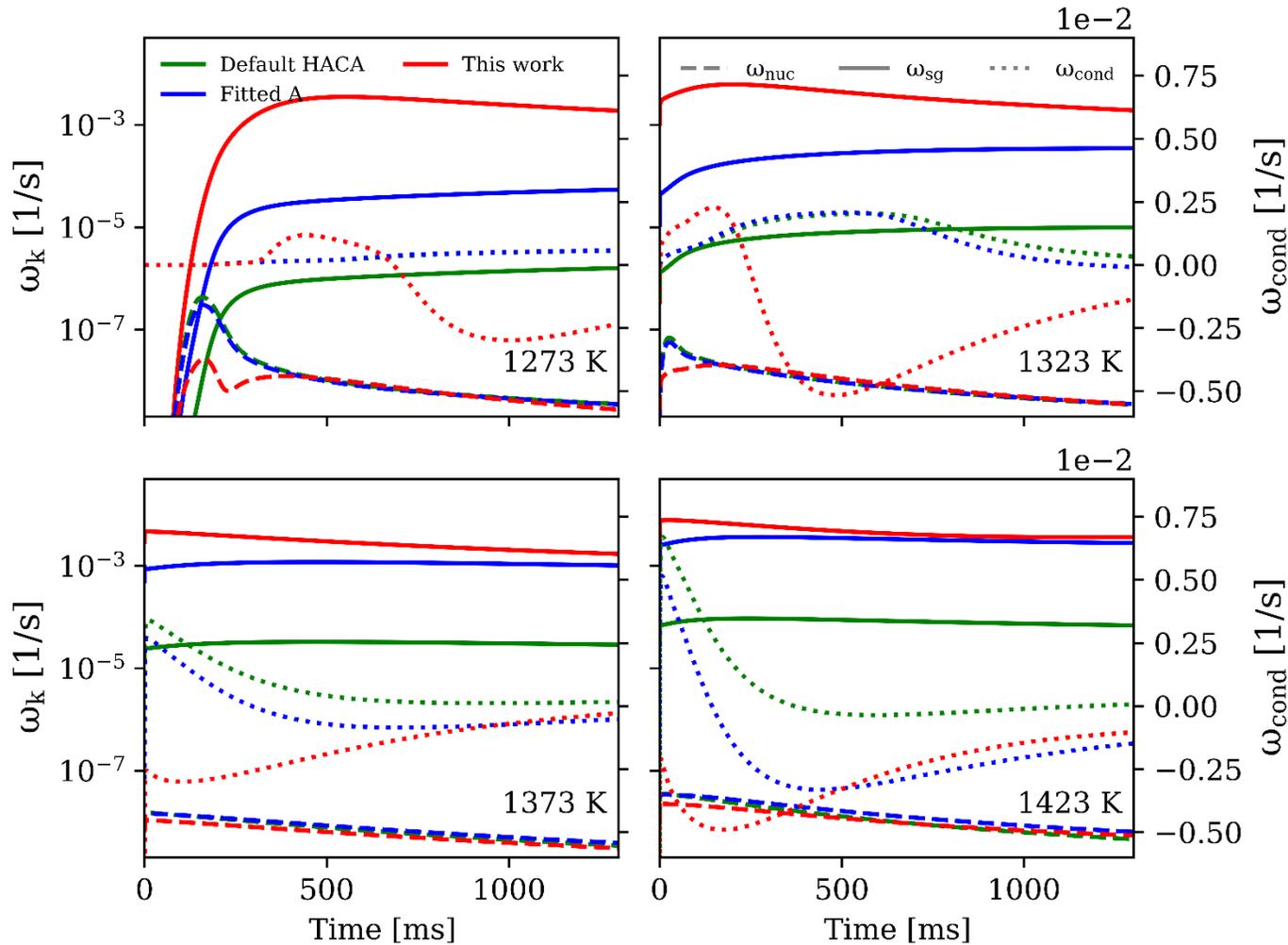
[1] Appel, J., Bockhorn, H., & Frenklach, M. (2000). Kinetic modeling of soot formation with detailed chemistry and physics: laminar premixed flames of C2 hydrocarbons. Combustion and flame, 121(1-2), 122-136.

[2] Hwang, J. Y., & Chung, S. H. (2001). Growth of soot particles in counterflow diffusion flames of ethylene. Combustion and flame, 125(1-2), 752-762.

Experimental and modeling reference (Ref.) results from:

Sánchez, Nazly E. Estudio de la formación de hidrocarburos aromáticos policíclicos (hap) en la pirólisis de acetileno y etileno. Diss. Universidad de Zaragoza, 2014.

PFR RESULTS



- Soot source terms for nucleation (ω_{nuc}), surface growth (ω_{sg}) and condensation (ω_{cond})

- Minimum ω_{sg} level required to induce a difference in the amount of soot produced

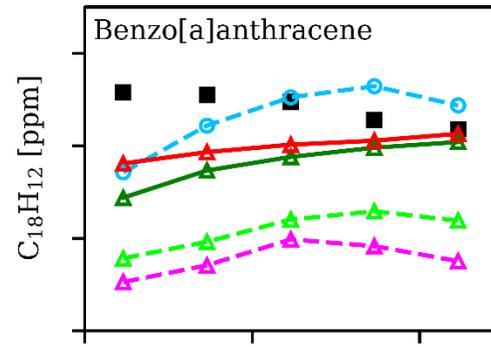
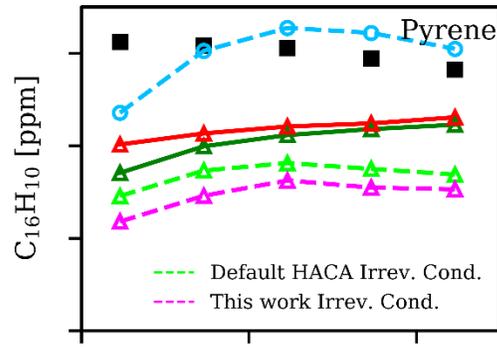
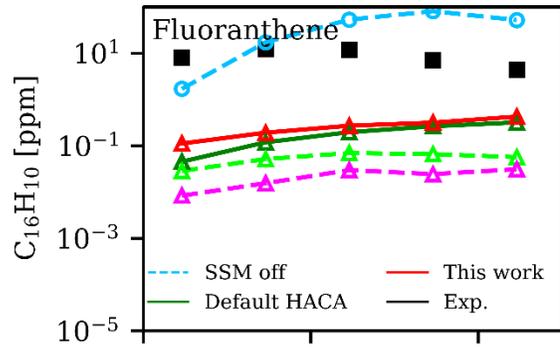
- Negligible effect on nucleation

- Increased relevance of condensation reversibility with the increase of surface growth

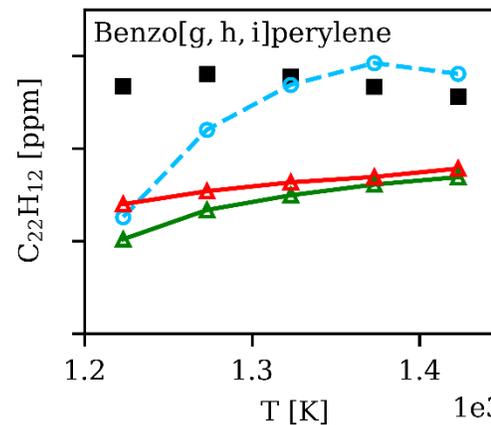
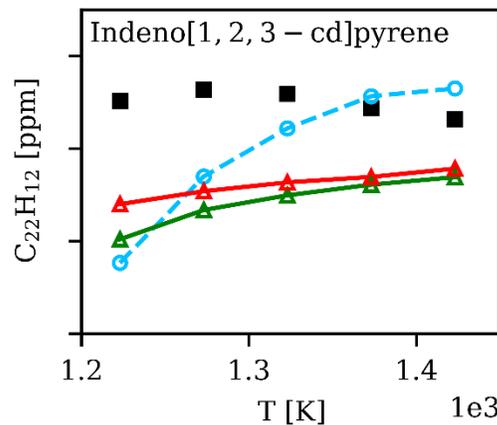
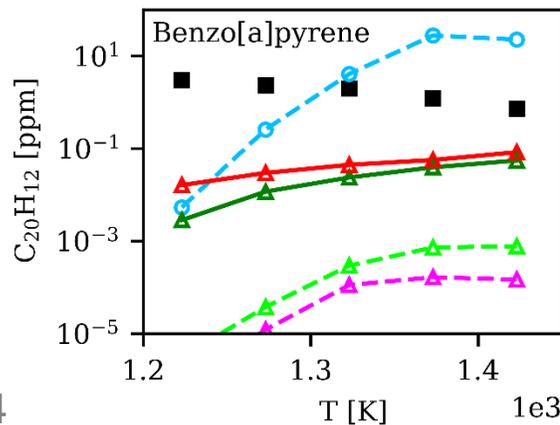
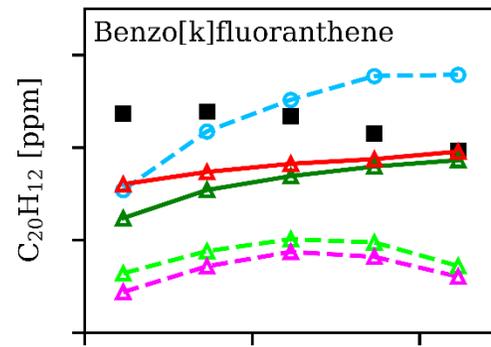
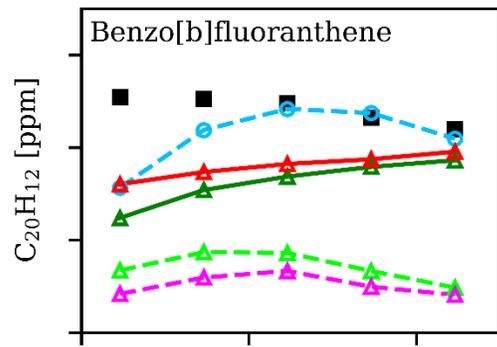
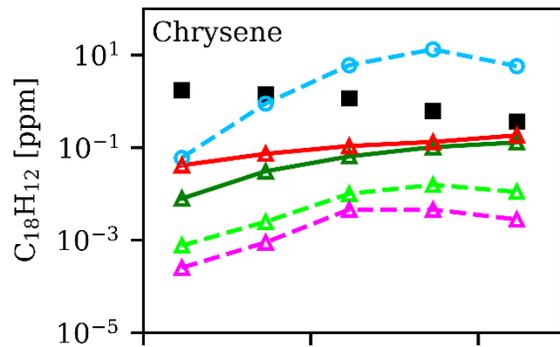
PFR RESULTS

Gas-phase mechanism

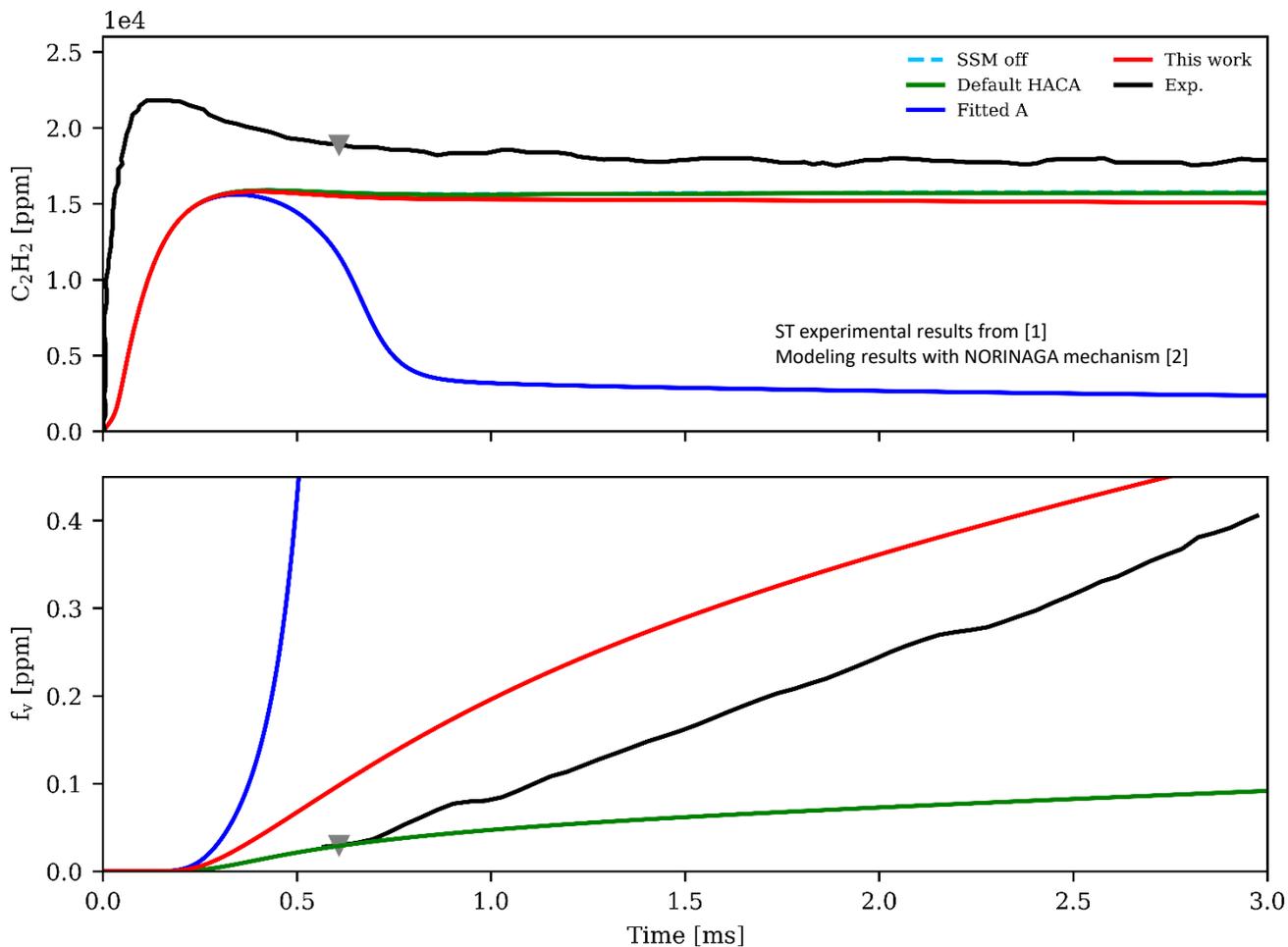
Soot surface growth mechanism



- General lower PAH concentration when soot modelling is accounted for (as expected)
- Larger discrepancies with experimental data without condensation reversibility



ST RESULTS



Experimental results from:
 Utsav, K. C., Mohamed Beshir, and Aamir Farooq. Proceedings of the Combustion Institute 36.1 (2017): 833-840.

Gas-phase mechanism

Soot surface growth mechanism

- Reference results without soot modeling
 - Slower initial increase in C_2H_2 mole fraction profile
- Setup 1: Soot volume fraction (f_v) under-prediction in line with PFR results
- Setup 2: Fitted A for PFR case over-estimates C_2H_2 addition
- Setup 3: Reasonable results despite under-prediction of soot onset

	Setup 1	Setup 2	Setup 3
A	80	2500	80
Surface growth mechanism	$C_{soot} + H \rightleftharpoons C_{soot}^* + H_2$ $C_{soot}^* + H \rightarrow C_{soot}$ $C_{soot}^* + C_2H_2 \rightarrow C_{soot} + H$	HACA + $C_{soot} + C_2H \rightarrow C_{soot}^* + C_2H_2$ $C_{soot} + CH_3 \rightarrow C_{soot}^* + CH_4$ $C_{soot} + C_3H_3 \rightarrow C_{soot}^* + AC_3H_4$	

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SUMMARY

- PAH results were not conclusive for the choice of the gas-phase mechanisms → Assessment complemented by fuel and acetylene concentration predictions
- Acetylene and soot predictions emphasize the importance of capturing the carbon balance between soot and the gas-phase → Soot surface growth mechanism is crucial to that end
- Additional steps added to the HACA mechanism [1,2] proved to enhance results both in the PFR (1.01×10^5 Pa, 1223 – 1423 K) and ST (3.80×10^5 Pa, 2179 K) cases
- Condensation reversibility improve PAH agreement with experimental data although large discrepancies still remain

[1] Appel, J., Bockhorn, H., & Frenklach, M. (2000). Kinetic modeling of soot formation with detailed chemistry and physics: laminar premixed flames of C2 hydrocarbons. *Combustion and flame*, 121(1-2), 122-136.

[2] Hwang, J. Y., & Chung, S. H. (2001). Growth of soot particles in counterflow diffusion flames of ethylene. *Combustion and flame*, 125(1-2), 752-762.

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