

## Water adsorption on silica nanoparticles

Keskinen Helmi<sup>1</sup>, Romakkaniemi Sami<sup>1</sup>, Jaatinen Antti<sup>1</sup>, Miettinen Pasi<sup>1</sup>, Joutsensaari Jorma<sup>1</sup>, Smith James N.<sup>1,2</sup> and Laaksonen Ari<sup>1,3</sup>

<sup>1</sup>Dept. of Physics and Mathematics, University of Eastern Finland, P.O. Box 1627, 70211 Kuopio, Finland.<sup>2</sup>National Center for Atmospheric Research, 80307, Boulder, CO USA

<sup>3</sup>Finnish Meteorological Institute, 00101, Helsinki, Finland

Keywords: fumed SiO<sub>2</sub>, nanoparticles, hygroscopic growth, water monolayers

The atmosphere contains small nanoparticles (below 100 nm) with concentrations from few to hundreds of thousands per cubic centimeter. These nanoparticles are exposed to water vapor in atmosphere (typical relative humidity (RH) range from 30 to 100 %) that produces films on their surfaces. This water layer on nanoaerosol has major effect as it affect the scattering and absorbance of radiation. It can also have important consequences for heterogeneous chemistry in the atmosphere as the kinetics of many reactions is strongly dependent on the phase of aerosol particles (Knipping et al. 2000). The sources of the atmospheric nanoparticles are natural, e.g., mineral dust, volcanic ash and sea spray (Satheesh and Krishna Moorthy, 2005; Kanakidou et al., 2005), or human made, e.g., aerosols produced in industrial processes, biomass burning and in fossil fuel combustion (Dubovik et al., 2001).

Hygroscopic tandem differential mobility analyzer (H-TDMA) (Liu et al., 1978) is ideal for nanoparticles hygroscopic growth detection as it has capacity to measure on-line the particle diameter (nano-DMA from 3 nm) in dry and humid conditions. Figure 1 shows the schematic of the H-TDMA set-up used in this study. The hygroscopic growth factors (GF) can be then defined by the relation of particle size in humid and dry air and when studying first monolayers adsorption on these nanoparticles the GF's need to be defined at low RH's. In fact, a water monolayer thickness is 0.19 nm and it would result the hygroscopic growth factor of 1.004 on the particle having diameter of 100 nm and 1.04 on 10 nm, respectively. Therefore, the detecting of the monolayers based on the size change on the larger sizes is very tricky as even the minimum measurement uncertainties would shade the growth. Here, the aim is to study the hygroscopic growth factors and water monolayers adsorption on fumed-SiO<sub>2</sub> nanoparticles (10-50 nm) at sub saturated water vapor conditions by nano-H-TDMA.

These preliminary experiments showed that for SiO<sub>2</sub> nanoparticles the GFs dropped strongly at 30 nm as seen from Figure 2a. This is most probably caused by agglomerates restructuring during humidification. Figure 2b shows the water monolayer (0.19 nm) number versus RH at selected dry particle diameters 8 and 10 nm for fumed SiO<sub>2</sub>. As expected, the monolayers number increase with RH and particle size. In fact, SiO<sub>2</sub> is hydrophilic and interact strongly with the water (Saliba et al., 2001). It is proposed that first 2 - 3 monolayers on SiO<sub>2</sub> interact with surface via hydrogen bonding with silanol groups (Verdaguer et al., 2007). The monolayers numbers defined by physisorption methods of water adsorption on flat SiO<sub>2</sub> surfaces (Figure 5,

(broken line); Verdaguer et al., 2007) are very close the ones observed here (Figure 5, spheres and triangles).

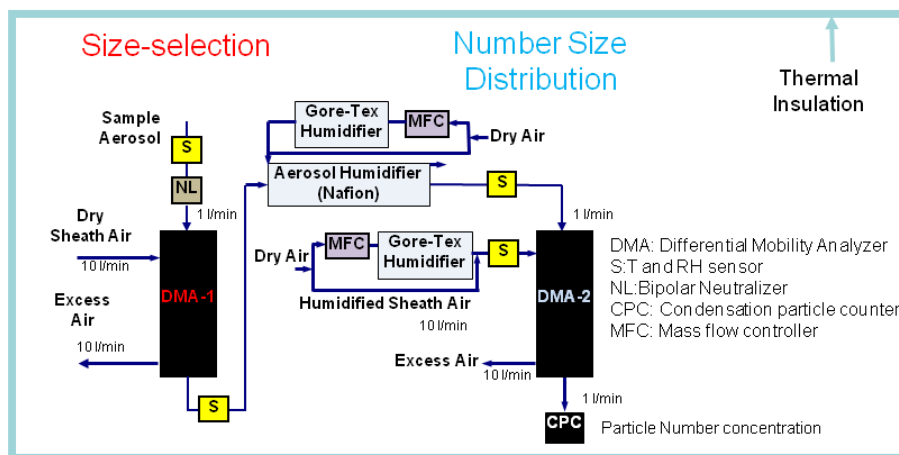


Figure 1 H-TDMA setup, first the aerosol was size selected by DMA-1 after that it was humidified in aerosol humidifier to desired RH (10 - 90%) and the humidified size distribution was measured by DMA-2 and condensation particle counter (CPC).

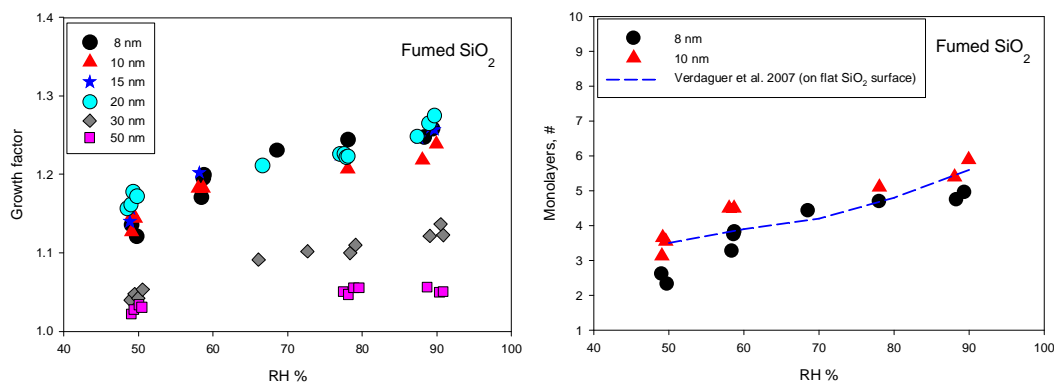


Figure 2 (a) Hygroscopic growth factor versus RH at selected dry particle diameters 8, 10, 15, 20, 30 and 50 nm for fumed  $\text{SiO}_2$  and (b) measured monolayer number versus RH at selected dry particle diameters 8 (spheres), 10 (triangles) nm for fumed  $\text{SiO}_2$  and for flat  $\text{SiO}_2$  surface (adapted from Verdaguer *et al.*, 2007).

Dubovik et al. (2002) Variability of absorption and optical properties of key aerosol types observed in worldwide locations, *J. Atmos. Sci.* 59, 590-608.  
 Kanakidou et al. (2005), Organic aerosol and global climate modeling: a review, *Atmos. Chem. Phys.*, 5, 1053-1123.  
 Knipping et al. (2000) Experiments and Simulations of Ion-Enhanced Interfacial Chemistry on Aqueous NaCl Aerosols, *Science*, 288, 301-306.  
 Liu et al. (1978) Aerosol mobility chromatograph—new detector for sulphuric-acid aerosols. *Atmos. Environ.* 12, 99–104.  
 Satheesh, S.K. and Krishna Moorthy, K. (2005) Radiative effects of natural aerosols: a review, *Atm. Environ.*, 39, 2089-2110.  
 Verdaguer et al., (2007) Growth and structure of water on  $\text{SiO}_2$  films on Si investigated by Kelvin probe microscopy and in situ X-ray Spectroscopies, *Langmuir*, 23, 9699-9703.



# Water adsorption on silica nanoparticles

Helmi Keskinen<sup>1</sup>, Antti Jaatinen<sup>1</sup>, Pasi Miettinen<sup>1</sup>, Sami Romakkaniemi<sup>1</sup>, Jorma Joutsensaari<sup>1</sup>, James N. Smith<sup>1,2</sup> and Ari Laaksonen<sup>1,3</sup>

<sup>1</sup>Dept. of Physics and Mathematics, University of Eastern Finland, P.O. Box 1627, 70211 Kuopio, Finland.

<sup>2</sup>National Center for Atmospheric Research, P.O. Box 3000, 80307, Boulder, CO USA

<sup>3</sup>Finnish Meteorological Institute, P.O. Box 503, 00101, Helsinki, Finland

## Motivation

### Silica is everywhere:

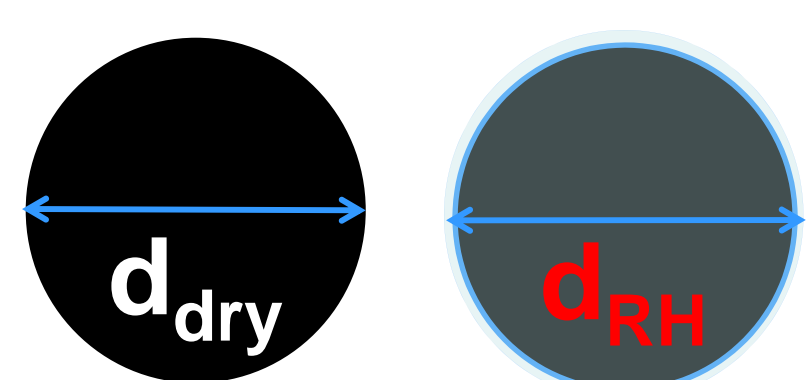
- **Natural sources:** desert dust (~80 % SiO<sub>2</sub>), volcanic ash, biomass burning
- **Anthropogenic sources:** additive (e.g., cosmetics and food industry). E.g., nanosized fumed silica is produced more than 1 million tons/year

### Water is everywhere:

- **Atmosphere:** nanoparticles impact on clouds and radiation
- **Breath:** nanoparticles solubility and surface properties impact on their cell interactions

## Methods

### Water on nanoparticles:

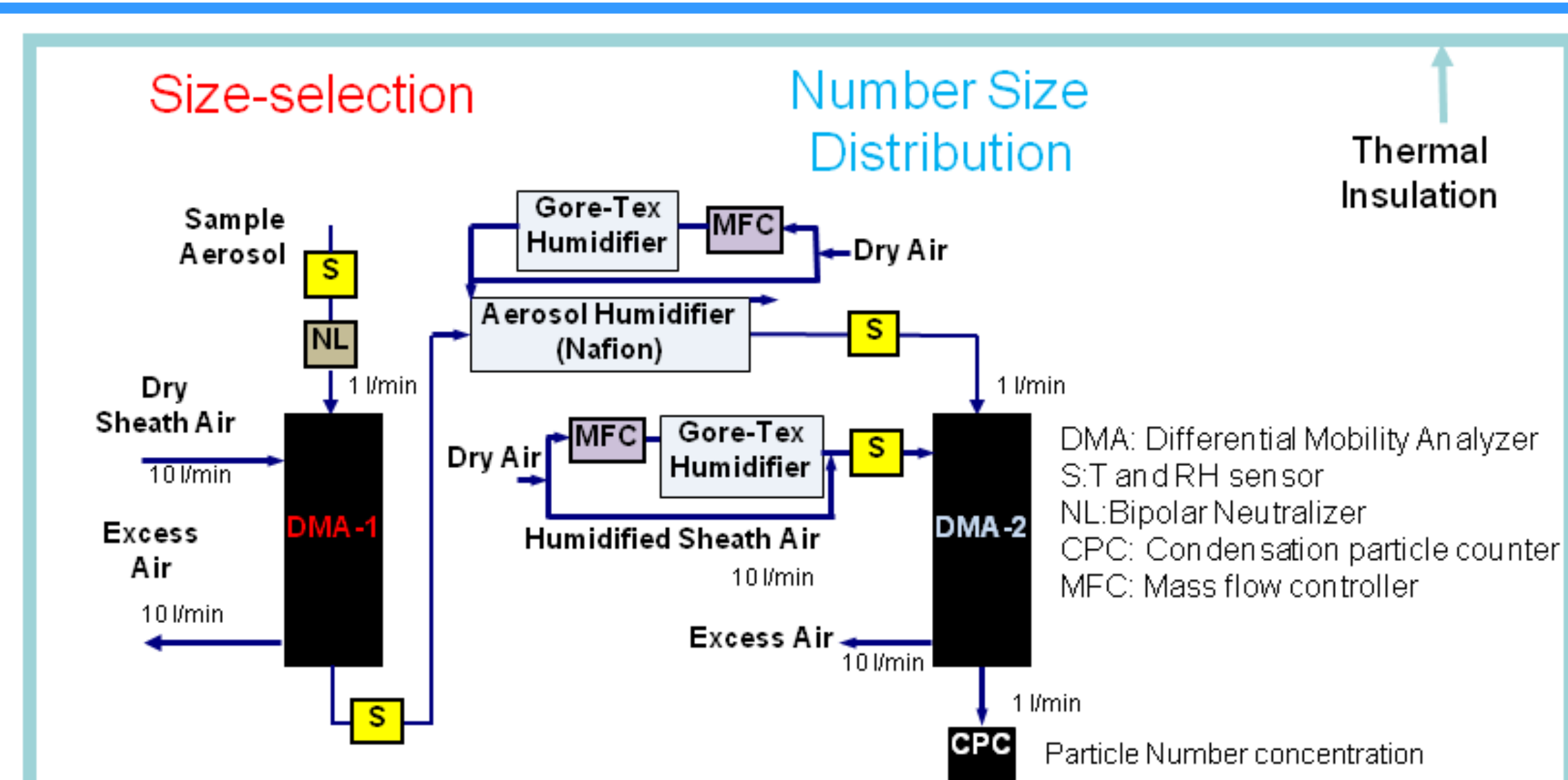


$$\text{Growth Factor (GF)} = d_{RH}/d_{dry}$$

$$\text{Number of monolayers} = (d_{RH} - d_{dry})/2l_{ml}$$

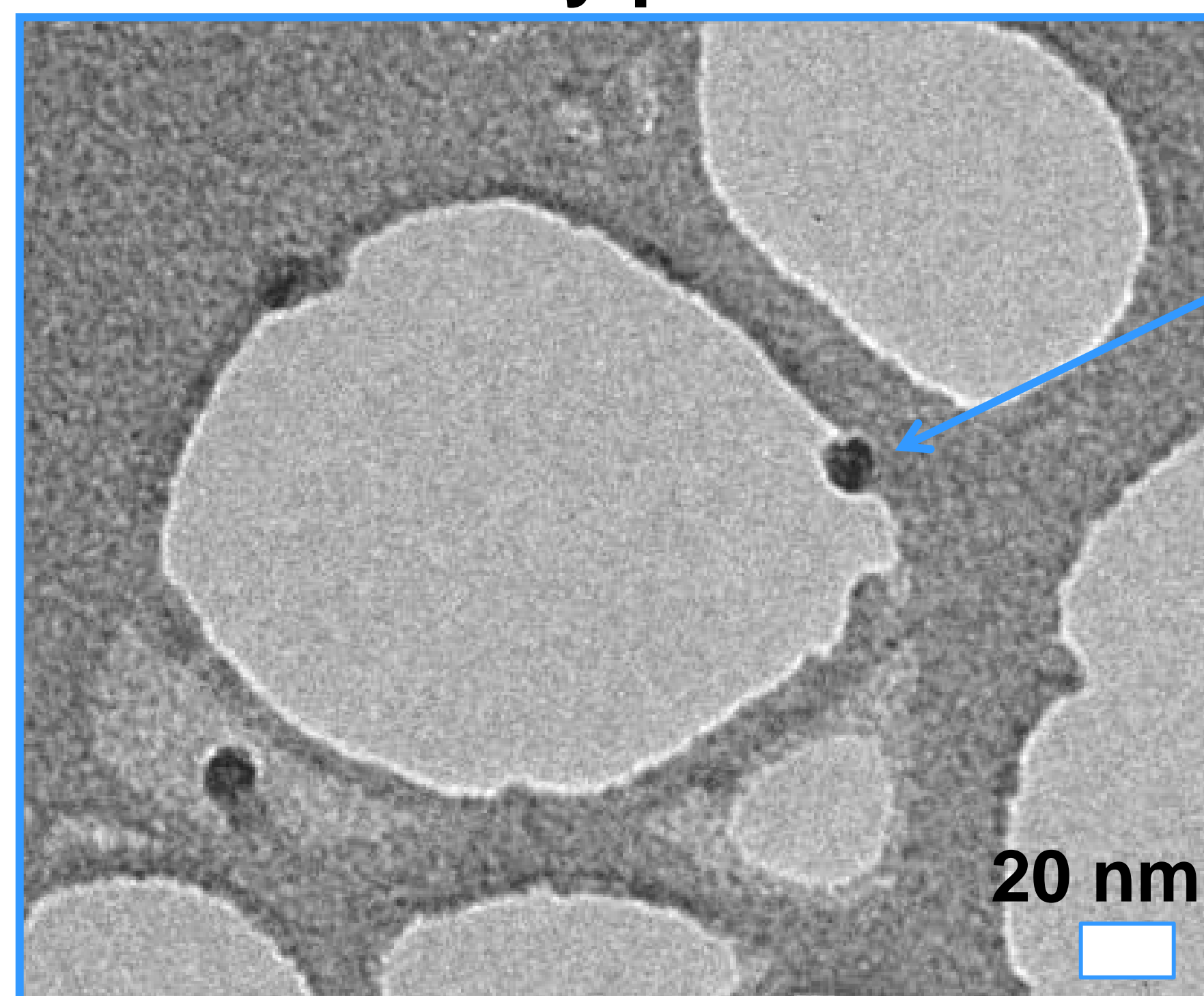
Water monolayer thickness is estimated to be 1.9Å

### Atomized fumed-silica ( $d_{prim} = 7$ nm) measured by: Hygroscopic Tandem Differential Mobility Analyzer



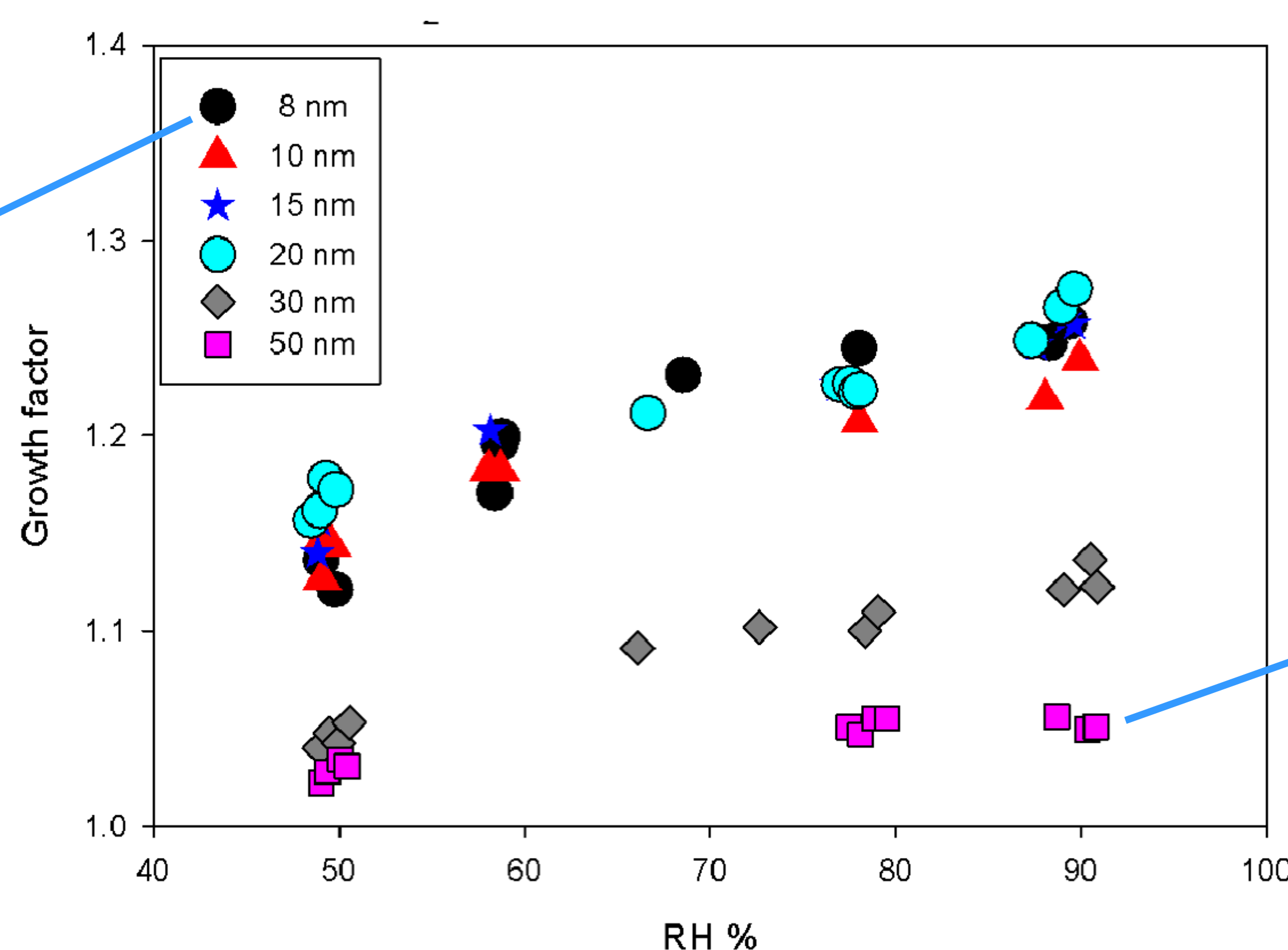
## Results

### Primary particles



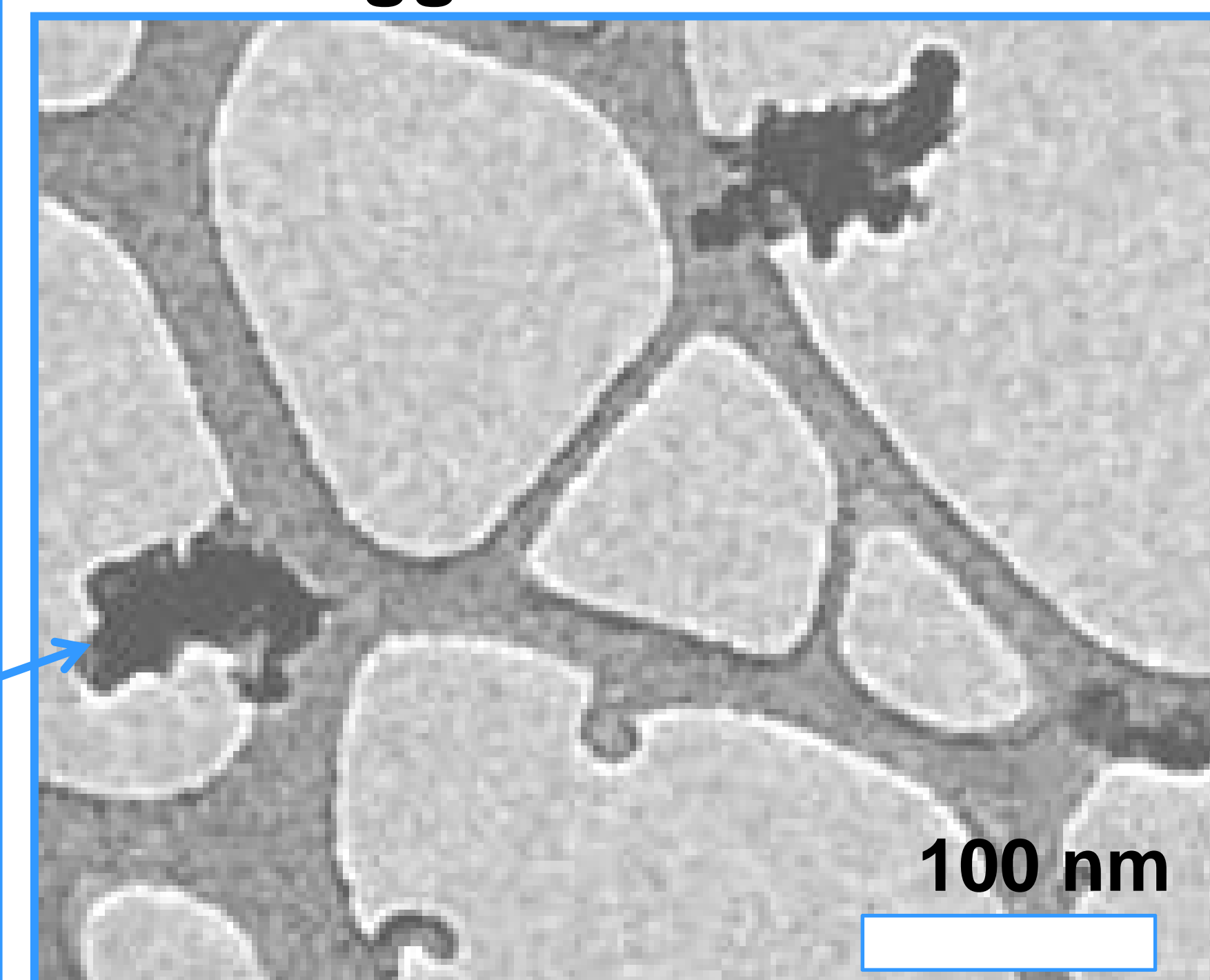
**Figure 1.** Transmission Electron Microscopy (TEM) photo of smaller primary particles.

### Growth factors



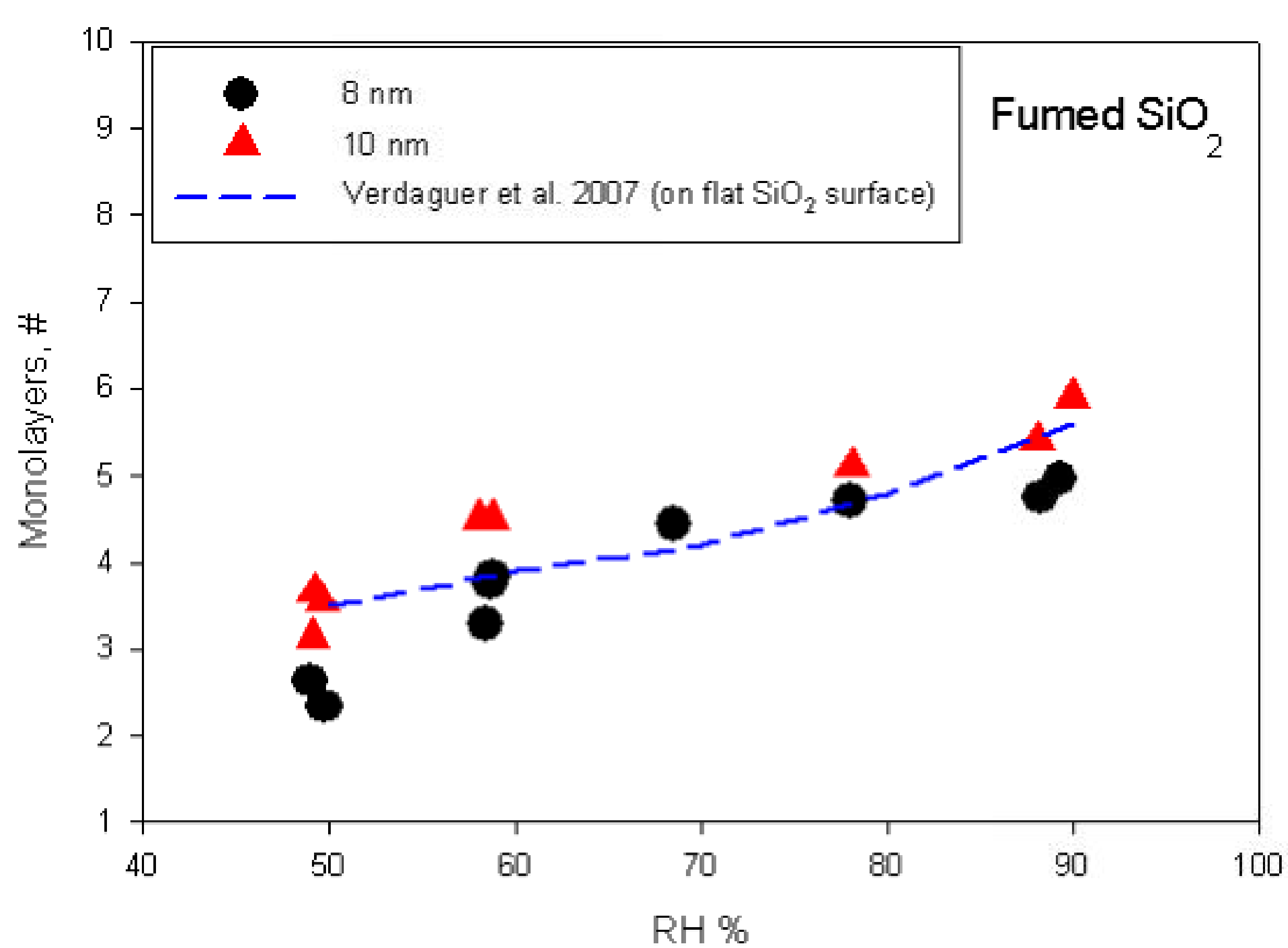
**Figure 2.** GF versus relative humidity (RH) at selected dry particle diameters 8, 10, 15, 20, 30 and 50 nm for fumed SiO<sub>2</sub>.

### Agglomerates



**Figure 3.** TEM photo of larger agglomerates.

### Number of water monolayers on SiO<sub>2</sub> particles



**Figure 4.** Monolayer number versus RH at selected dry particle diameters 8 (spheres), 10 (triangles) nm for fumed SiO<sub>2</sub> and for flat SiO<sub>2</sub> surface (adapted from Verdaguer *et al.*, 2007).

## Conclusions

- Size-selected 8 -10 nm particles were spherical
- From larger size fractions also agglomerates were detected
- Growth factors for agglomerates dropped (→restructuring)
- at RH = 55-90% 4 to 6 water layers were detected on 8 and 10 nm particles
- Water monolayer number agreed well with literature

## Acknowledgements

This work was supported by the academy of Finland (Center of Excellence, No 1118615 and decision numbers 123466, 118230), Kone foundation and Saastamoinen foundation.

## References

Verdaguer *et al.*, (2007) Growth and structure of water on SiO<sub>2</sub> films on Si investigated by Kelvin probe microscopy and in situ X-ray Spectroscopies, *Langmuir*, 23, 9699-9703.