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Nanoparticle formation in modern Diesel vehicle exhaust:

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Nanoparticle formation in modern Diesel vehicle exhaust: New insights from innovative exhaust measurements of key precursor gases

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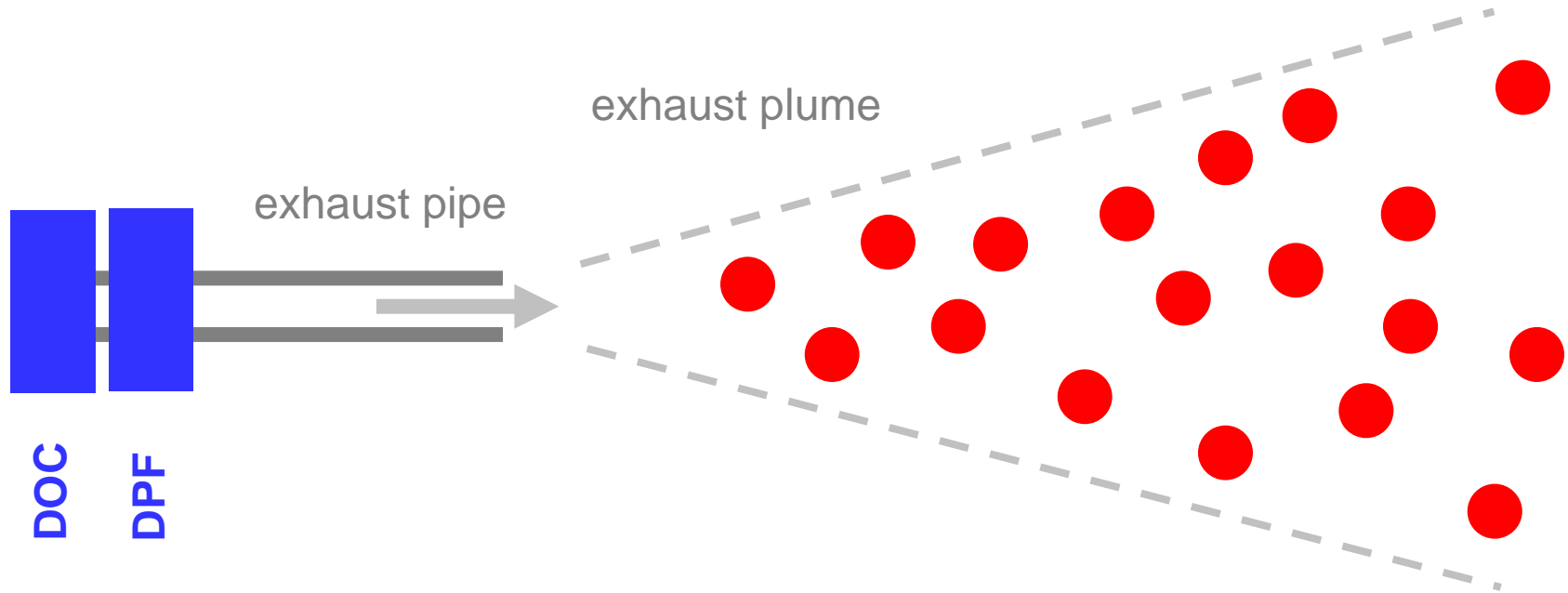
For more information see

paper in preparation

contact:

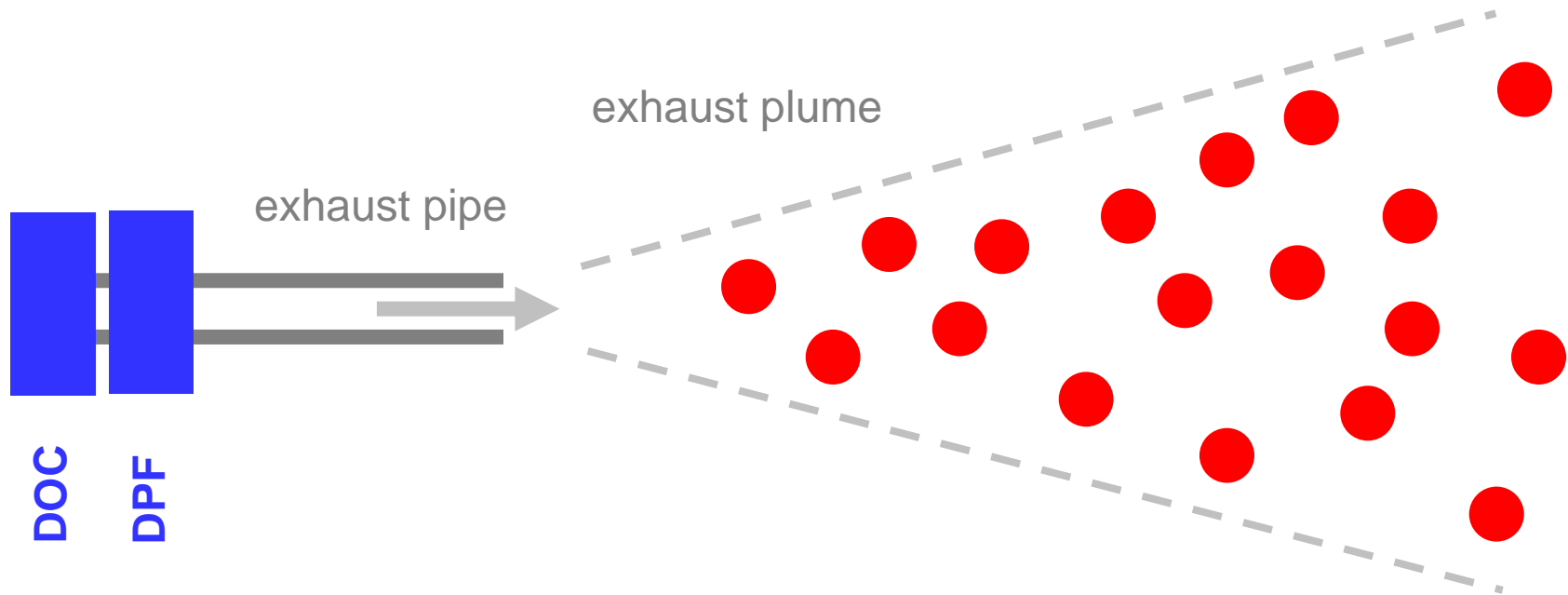
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Modern Diesel vehicles equipped with **after treatment systems (DOC + DPF)**



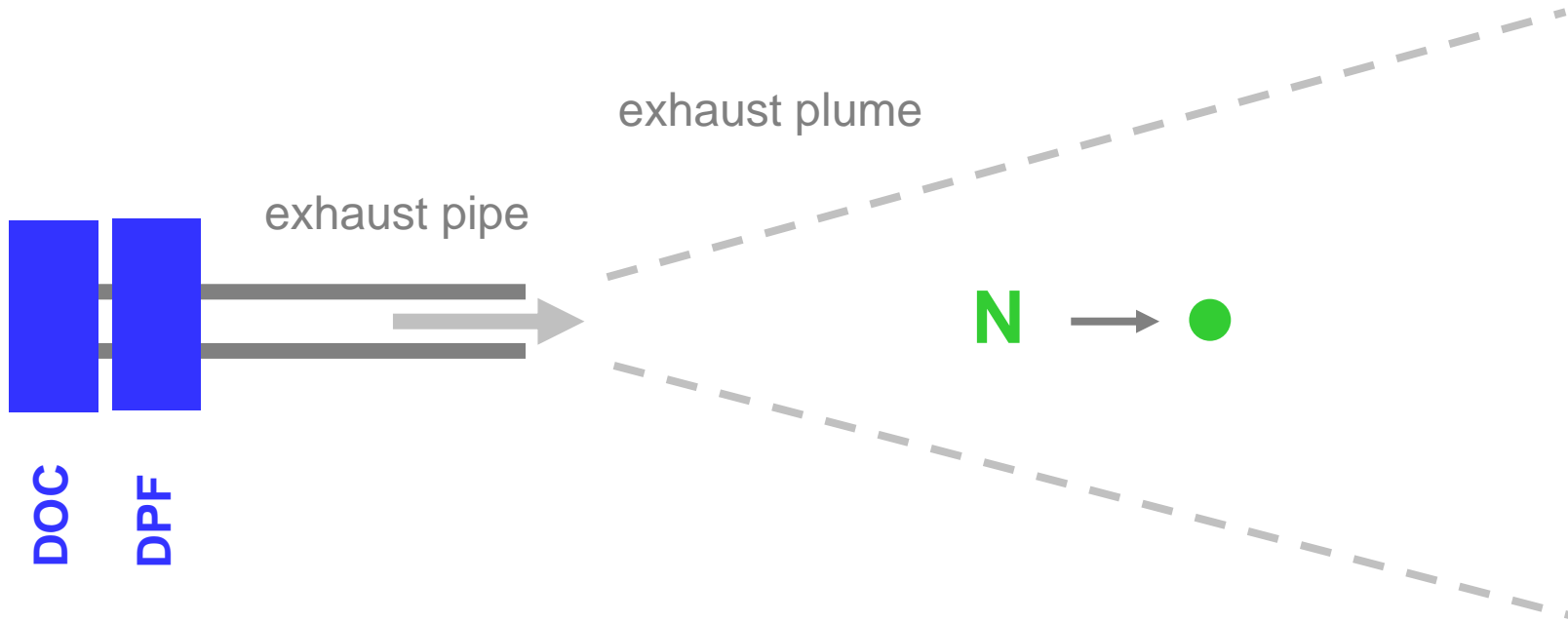
- Semi-volatile particles ● (D=10 nm) are present in large concentrations

Diesel exhaust : **with** after treatment (DOC + DPF)

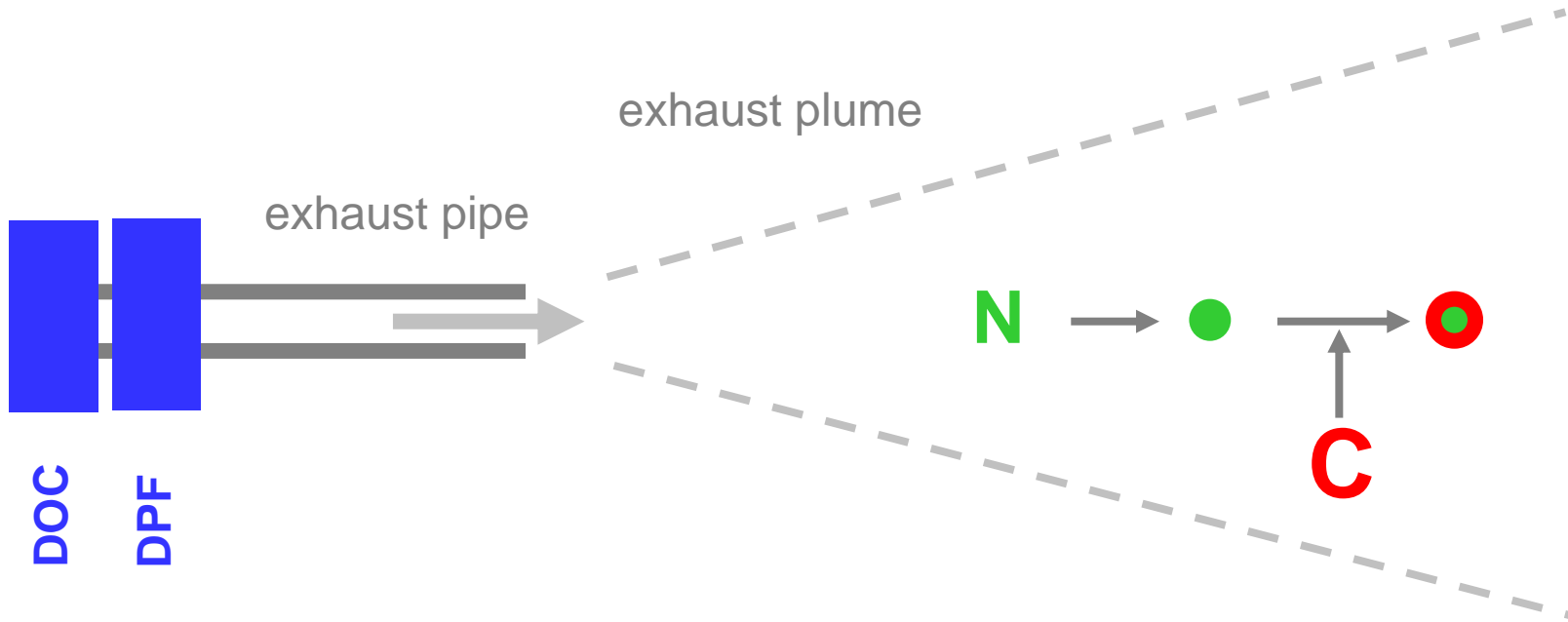


- Semi-volatile particles ● (D=10 nm) are present in large concentrations
- Must be formed downstream of DPF
- Requires presence of nucleating gases (**N**) and condensing gases (**C**)

Diesel exhaust : **with** after treatment (DOC + DPF)



Diesel exhaust : **with** after treatment (DOC + DPF)



Nucleation particles (NUP)

- Mechanism of **formation** and **chemical nature** only poorly understood
- NUP **precursor gases** not known

Nucleation particles (NUP)

- Mechanism of **formation** and **chemical nature** only poorly understood
- NUP **precursor gases** not known
- **di-acids** are conceivable candidates
(have low saturation vapor pressures due to efficient hydrogen-bonding)

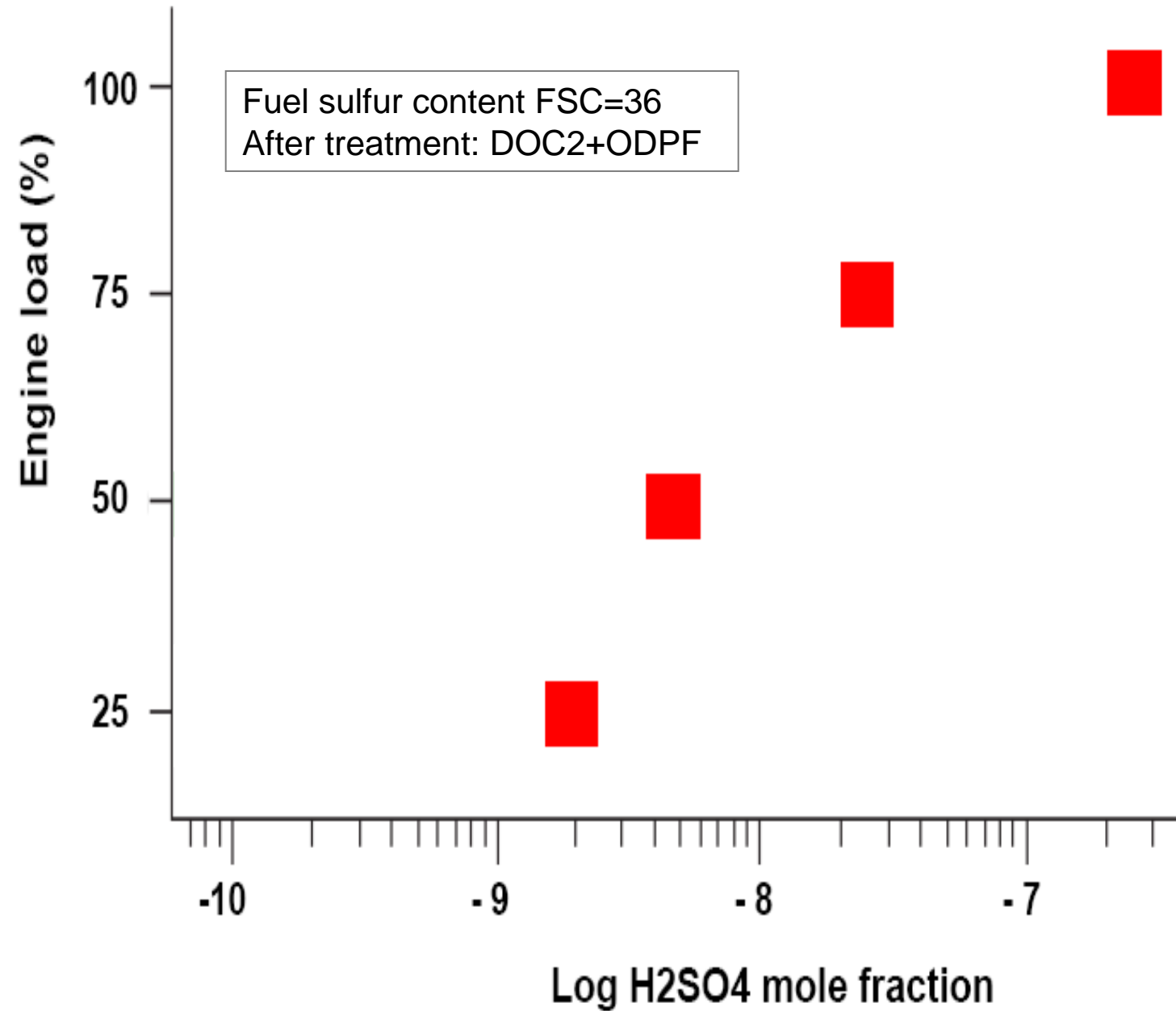
Experiments at MAN engine test lab (Nuernberg)

- **NUP precursor gas:** measurements (acids)
- **Measurement method:** **IMR-ITMS** (developed by *MPIK Heidelberg*)
- **On-line** : heated exhaust
- **Off-line** : sampling on stainless steel surface, followed by thermodesorption
- Heavy duty Euro 4 Diesel vehicle **engine**:
 - MAN 323 kW , 6 cylinder turbo charged common rail
 - displacement: 10.6 l, torque: 2220 Nm
- **Parameters** varied during measurements : **FSC** , **ATS** , **EL**

On-line measurements

Sulfuric acid

Example of an on-line measurement of gas-phase H₂SO₄ in heated Diesel exhaust



H₂SO₄ formation increases with:

- Fuel sulfur content **FSC**
- Engine load **EL**
- Aftertreatment system **ATS**

„Fuel sulfur conversion fraction F“

- Diesel engine without ATS : F < 1%
- Diesel engine with ATS : F < 20-30 %

Comparison: **Air craft exhaust plumes**

Sulfur conversion efficiency

„Fuel sulfur conversion fraction F“

- Diesel without ATS : about 1%
- Diesel with ATS : about 20-30 %
- Jet air craft : about 2-4 %

Diesel vehicle exhaust:

Di-carboxylic acids

Di-carboxylic acids

- Various DCA have been observed
- DCA are correlated with H₂SO₄

Off-line measurements

- **Sampling**
of exhaust components on heated stainless steel surface
- **Thermodesorption**
stepwise increase of temperature (to 420 C)
- **IMR-ITMS measurement**
of desorbed gases

Example of sampler measurement

Sampler 16

NI-CIMS

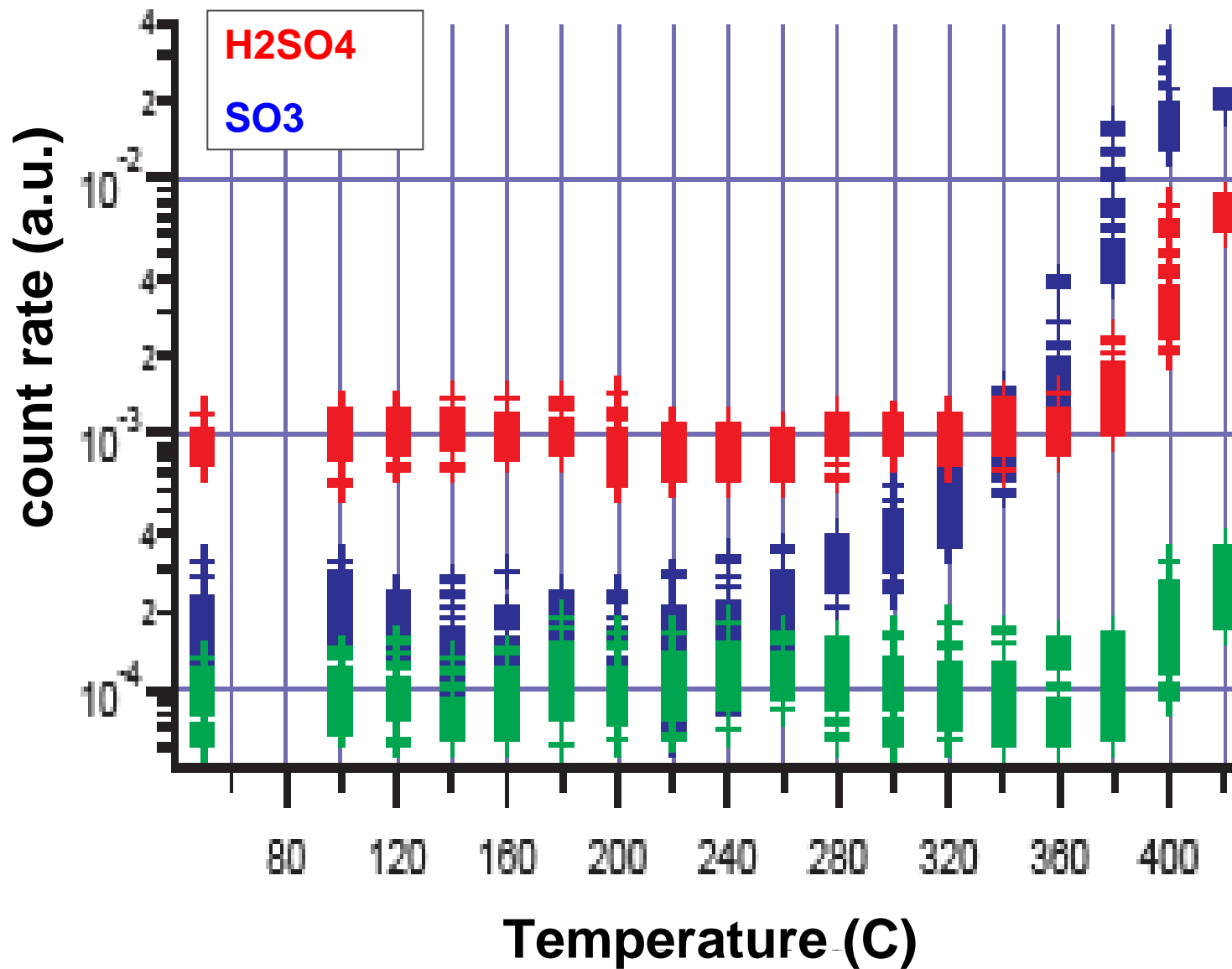
Sampling-T: 50 C

Thermodesorption: - 400 C

FSC=6 ppmM

no ATS !

Example of an off-line measurement of gas-phase H₂SO₄ and SO₃



Thermodesorption of H₂SO₄

- H₂SO₄ condensate is more stable than H₂SO₄/H₂O condensate
- H₂SO₄ condensate does not seem to be ammonium sulfate ! (measured desorbed NH₃ is much less than H₂SO₄)

Di-carboxylic acids

Relative abundances of
thermodesorbed acids
(for $T = 120\text{ C}$)

Ion Identification

- High precision **mass** measurements
- **Fragmentation** studies of mass selected ions:
 - for different collision **energies**
 - for different collision **gas atoms** (He , Ar)

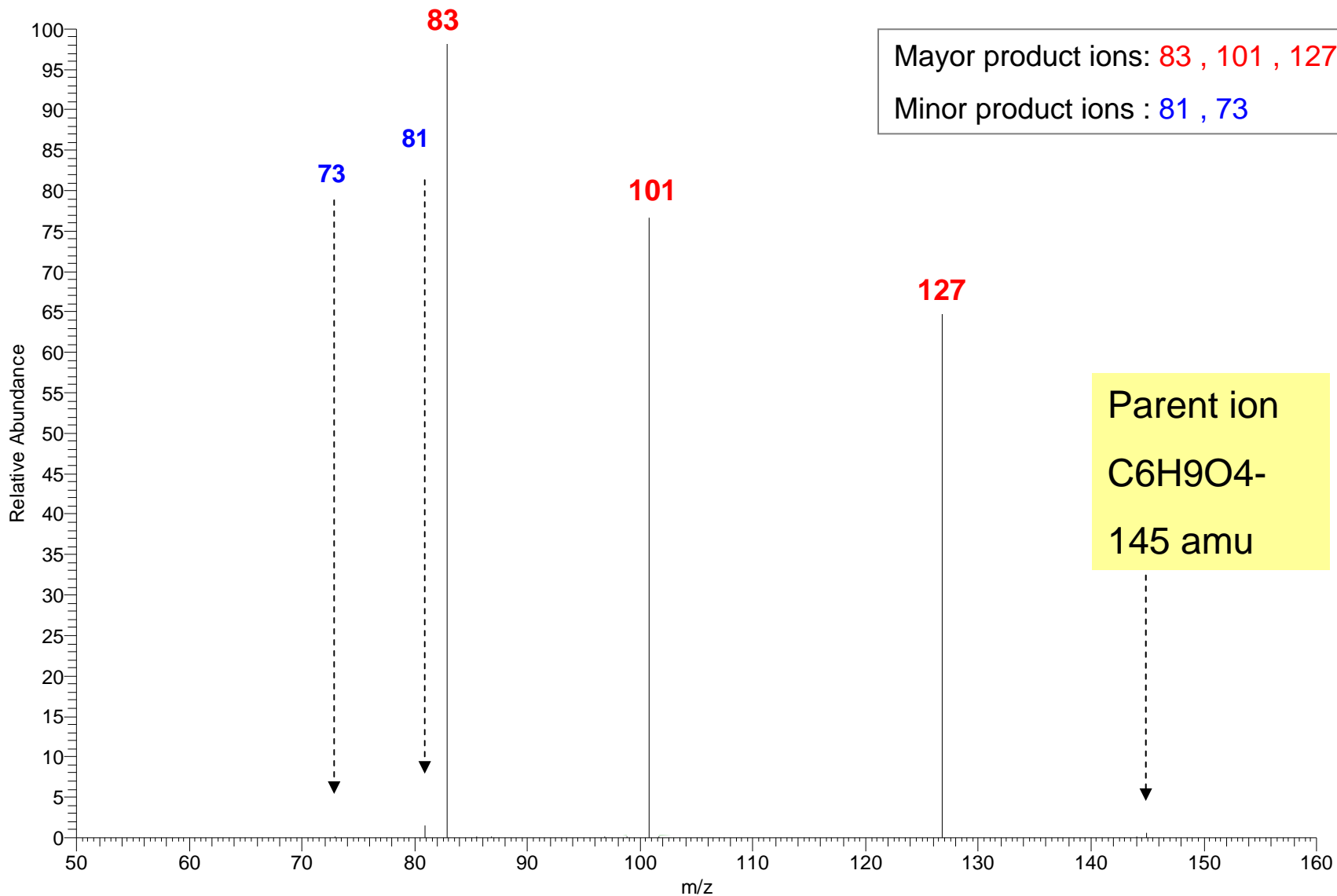
Fragmentation studies

Example:

Adipic acid $C_6H_{10}O_4$

Adipic acid

MS-MS mass spectrum: parent ion $C_6H_9O_4^-$; collision gas He ; CE = 30 eV



Aerosol formation and growth

Conclusions

- H₂SO₄ increases with **EL**
- H₂SO₄ increases with **FSC**
- H₂SO₄ increases with **ATS**
- Strong H₂SO₄ **store and release** effects

Summary and Conclusions

- H₂SO₄ increases with **EL**
- H₂SO₄ increases with **FSC**
- H₂SO₄ increases with **ATS**
- Strong H₂SO₄ **store and release** effects
- **Organic acids** correlated with H₂SO₄
- NUP **conc.** increases with H₂SO₄
- NUP **diameter** increases with H₂SO₄
- NUP **volume conc.** increases with H₂SO₄
- NUP occasionally correlated with **acids other than H₂SO₄**

- For more information see paper in preparation (contact: **frank.arnold@mpi-hd.mpg.de**)

ThankYou
for your interest