

## Two-nozzle flame synthesis of Pt/Ba/Al<sub>2</sub>O<sub>3</sub> for NO<sub>x</sub> storage

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A novel two-nozzle flame spray pyrolysis (FSP) process is presented for one-step preparation of Pt/Ba/Al<sub>2</sub>O<sub>3</sub> particles as used for NO<sub>x</sub> storage-reduction (NSR) catalysts. This material is of particular interest for engines operating under lean conditions for the NO<sub>x</sub> abatement. According to the NSR concept, NO<sub>x</sub> is stored under lean conditions in the form of alkali or alkaline-earth nitrates (in particular Ba(NO<sub>3</sub>)<sub>2</sub>) and reduced over a noble metal into N<sub>2</sub> during fuel rich periods.

Recently it has been shown that different Ba phases of impregnated materials strongly affect the NO<sub>x</sub> storage capacity of Pt/Ba/Al<sub>2</sub>O<sub>3</sub>, and BaCO<sub>3</sub> decomposing at low temperatures (LT-BaCO<sub>3</sub>) has been identified as the most active Ba species in the NO<sub>x</sub> storage process.<sup>1</sup>

Flame aerosol and in particular flame spray technologies are versatile and continuous processes for production of a variety of ceramic nanoparticles. In contrast to spray pyrolysis, flame spray pyrolysis (FSP) is based on combustible precursor solutions, which provide the energy for the process.<sup>2,3</sup> A metal containing precursor solution is dispersed, ignited, and combusted. After evaporation and conversion of the metal precursor, particles are formed in the gas phase and supported noble metal catalysts (i.e., Pt/Al<sub>2</sub>O<sub>3</sub>) consisting of Pt particles (<5 nm) finely dispersed on Al<sub>2</sub>O<sub>3</sub> particles (10-40 nm) have been made by FSP.

Compared to the conventional single-nozzle setup during FSP, the present stereoscopic two-nozzle setup adds further flexibility for the control of important flame parameters, such as temperature and concentration fields, that affect particle formation, and affords the control of particle mixing at the nano-level in multicomponent systems. The use of two separate nozzles, one as aluminum and the other as a barium/platinum source, resulted in individual Al<sub>2</sub>O<sub>3</sub> and monoclinic BaCO<sub>3</sub> nanoparticles, exhibiting good NO<sub>x</sub> storage activity. In contrast, using a single-nozzle process resulted in Al<sub>2</sub>O<sub>3</sub> particles with amorphous Ba species with negligible NO<sub>x</sub> storage capacity. Increasing the inter-nozzle distance resulted in late mixing of the two flame products and increased the amount of crystalline BaCO<sub>3</sub>. At ambient conditions the as-prepared monoclinic BaCO<sub>3</sub> transformed into orthorhombic BaCO<sub>3</sub>. Independent of the Ba loading, flame made nano-crystalline BaCO<sub>3</sub> showed a low thermal stability (decomposition below 900 °C, LT-BaCO<sub>3</sub>) that was distinctly different from its "bulk" behavior (decomposition above 900 °C).

The synthesized materials were characterized by transmission electron microscopy, nitrogen adsorption, X-ray diffraction, and temperature programmed decomposition and tested for their NO<sub>x</sub> storage behavior.

(1) Piacentini, M.; Maciejewski, M.; Baiker, A. *Appl. Catal., B* **2005**, *60*, 265.

(2) Pratsinis, S. E. *Prog. Energy Combust. Sci.* **1998**, *24*, 197.

(3) Mädler, L. *KONA* **2004**, *22*, 107.

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## Objective

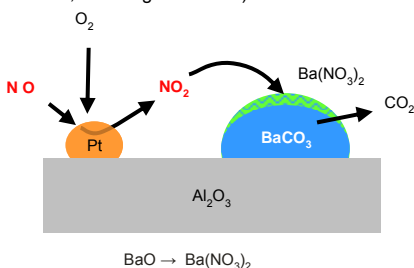
NO<sub>x</sub> storage reduction (NSR) catalysts are used for abatement of NO<sub>x</sub> from engines operating under lean conditions where conventional TWC catalysts are inefficient.

Here a flame process was used for synthesis of Pt/Ba/Al<sub>2</sub>O<sub>3</sub> NSR-catalyst. Flame synthesis is a scalable and continuous process for the synthesis of a variety of nano-particles, including noble metal catalysts. The two nozzle flame spray pyrolysis setup allows controlling particle mixing at the nano level in multicomponent systems. The structural properties of as-prepared materials were characterized and the catalyst have been tested for NO<sub>x</sub> storage.

## Working principle

### Storage cycle

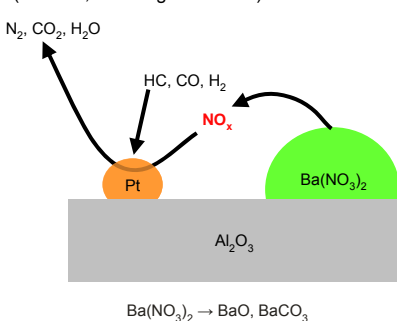
(fuel lean, oxidizing conditions)



During lean fuel condition effluent NO<sub>x</sub> is stored in the structure of alkali- or alkaline-earth nitrates. Here BaCO<sub>3</sub> is forming Ba(NO<sub>3</sub>)<sub>2</sub>. Additionally the noble metal catalyses the formation process.

### Reduction cycle

(fuel rich, reducing conditions)

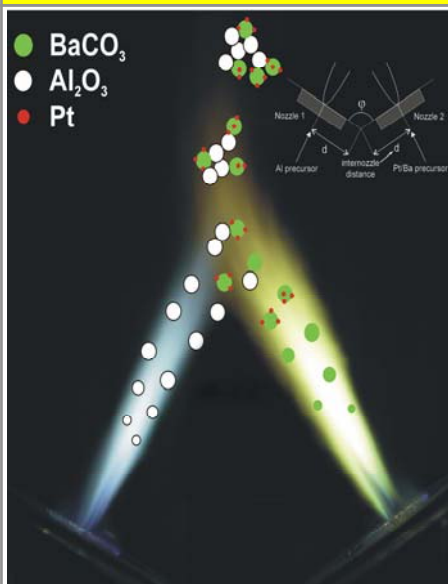


Is the capacity of the storage material exhausted the Ba will be regenerated using fuel rich conditions in the exhaust gas being provided by the engine or by extra fuel injection directly into the NSR catalyst

## Acknowledgements

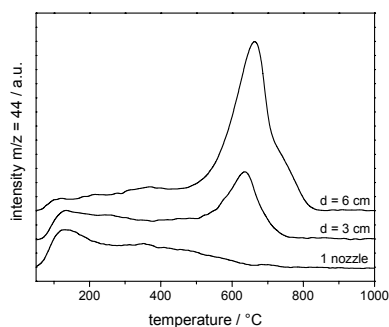
We would like to thank Dr. Frank Krumeich from ETH Zürich for the electron microscopy analysis and kindly acknowledge financial support by ETH Zürich (ETH Research Grant TH-2/03-2).

## Setup

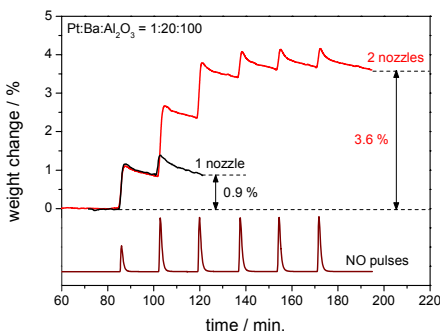


Two nozzle flame spray pyrolysis of Pt/Ba/Al<sub>2</sub>O<sub>3</sub> where Al and Ba precursor solutions are sprayed in two separated FSP nozzles. After the formation of individual Al<sub>2</sub>O<sub>3</sub> and Pt/BaCO<sub>3</sub> particles the two flames combine resulting in a well mixed powder.

## Low Temperature BaCO<sub>3</sub>



TPD CO<sub>2</sub> evolution profiles during decomposition of BaCO<sub>3</sub> for as prepared Pt/Ba/Al<sub>2</sub>O<sub>3</sub> from one or two nozzles.

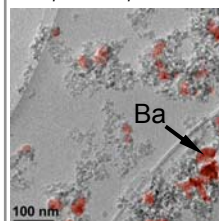


TG analysis during NO<sub>x</sub>-storage for flame made Pt/Ba/Al<sub>2</sub>O<sub>3</sub> made with one or two nozzles. NO pulses were injected into 5% O<sub>2</sub>/He. In contrast to the two-nozzle made catalyst, only little NO was stored on the powder prepared with one nozzle.

## BaCO<sub>3</sub> morphology

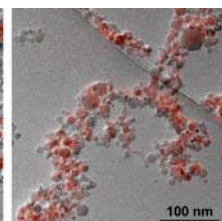
2 Nozzles:

Separate Ba-particles

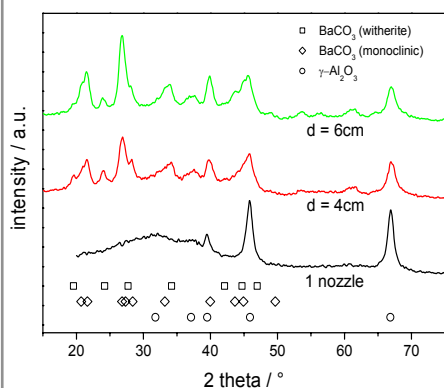


1 Nozzle:

Mixture of Ba-Al compounds



TEM images with corresponding Ba mappings (shown as a red overlay) for Pt/Ba/Al<sub>2</sub>O<sub>3</sub> made with one and two nozzles. The used two nozzles resulted in the formation of individual BaCO<sub>3</sub> particles, whereas Ba is distributed all over the Al<sub>2</sub>O<sub>3</sub> particles in the powder prepared with one nozzle.



XRD pattern of as-prepared Pt/Ba/Al<sub>2</sub>O<sub>3</sub> made with one and two nozzles at different inter-nozzle distances. Increasing the inter nozzle distance resulted in higher amounts of crystalline BaCO<sub>3</sub>. The initially formed monoclinic BaCO<sub>3</sub> phase transformed into the stable orthorhombic form within 4 weeks.

## Conclusions

A novel two-nozzle flame spray pyrolysis (FSP) process was developed for one step synthesis of BaCO<sub>3</sub> and Al<sub>2</sub>O<sub>3</sub> nanoparticles well-mixed at the nano level. The flame made BaCO<sub>3</sub> particles decomposed at low temperatures (LT-BaCO<sub>3</sub>) compared bulk BaCO<sub>3</sub> particles (600°C vs 1000°C).

NO pulse experiments revealed no NO<sub>x</sub> storage capacity for Pt/Ba/Al<sub>2</sub>O<sub>3</sub> made with one nozzle, but good storage for catalyst made with 2 nozzles.

## Further reading

- R. Strobel *et al.*, Chem. Mater. 18 (2006) 2532.
- R. Strobel *et al.*, Thermochem. Acta 445 (2006) 23.
- M. Piacentini *et al.*, J. Catal. 243 (2006) 43.
- R. Strobel *et al.*, J. Catal. 213 (2003) 296.
- M. Piacentini *et al.*, Appl. Cat. B Environ. 60 (2005) 265.
- M. Piacentini *et al.*, Top. catal. 30-31 (2004) 71.